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Heavy Metals Level Evaluation of Drinking Water Supply In Kirkuk City/NE of Iraq

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Abstract

An evaluation of pollution level has been done for drinking water which is used in Kirkuk city by heavy metals (As,Cd,Cu,Fe,Mn,Ni,Zn,Cr,Pb) samples were collected during wet and dry of 2016-2017 from Kirkuk unified water supplied project (WTP) which is supply the city with drinking water, as well as from water of tanks type (GST2), and also from (tap water) (Zone3). The results showed the (As,Cd,Cu,Fe,Mn,Ni,Zn,Cr,Pb)inppbfor concentration of the (WTP)are (0.5,0.6,6.45,38,4.6,2.5,6537,0.58,1.4) (0.6,0.8,6.76,46,5.5,3.5,6675,1,2.4) for (GST) (0.5,0.63,6.46,52.3,4.4,3.6,6550,0.6,2.5) (0.60, 0.7, 6.78, 63.7, 6.7, 3.7, 6680, 1.1, 2.6) and for tap water are (0.53,0.65,7.00,60.2,4.4,3.65,7200,0.8,2.7) (0.60, 0.71, 7.10, 67.6, 6.8, 3.75, 7320, 0.9, 2.75) for two seasons respectively the concentrations are within the allowed limits by WHO just Zn is exceed the allowed limits . the level of health risks were also assessed by using a number of indices (HQ)ing, (HQ) derm, (HI), (CR) ing (CR) derm for adult and children and for two seasons . (HI) value for Zn show health risks this high value of (HI) for Zn make water undrinkable.

Key words: Heavy metals, Drinking water pollution, (HI) Index, Kirkuk city, Iraq.

تقييم مستوى تلوث مياه الشرب بالعناصر الثقيلة في مدينة كركوك اشمال شرق العراق

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لاغلب العناصر ماعدا عنصر الخارصين حيث اظهرت قيم (HI) عالية وهذا يجعل المياه غير صالحة للشرب.

Introduction

Heavy metals are the general term encompassing the group of metals and non-metals which density is above 6 g/cm³ [1,2]Trace elements may come from natural sources, leached from rocks and soil according to their geochemical mobility or come from anthropogenic sources like industrial discharge, urban runoff, sewage effluents[3]. Heavy metals presence in nature usually is not dangerous for the environment because they are present only in very small quantities .and become polluted only if it's present in large quantities.. trace heavy metals analysis is an important part of public health studies [4], Some metals at trace levels in our metabolism are important for good health, but they become dangerous and toxic to human health if their concentrations rise from natural values, and cause many health risks like reduced growth, cancer, nervous system damage, and in extreme causes death. Drinking water contamination with heavy metals is one of the most important environmental issues as they are toxic even at low concentration [5].there are 23 of heavy metals: antimony, arsenic, chromium[6,7] gold, cerium, iron[8,9,10], platinum, cadmium[11,12] gallium, copper, bismuth, lead[13,14], manganese[15,16], silver, tellurium, thallium, tin, uranium, vanadium, nickel[17,18], mercury, and zinc the most important of them is mercury, cadmium, lead, copper[19], iron, etc.,. The current study aims to assess the level of heavy metals contamination in drinking water in Kirkuk city and for the first time, there are no researches touched to study the heavy metals in drinking water for Kirkuk city.

Study Area

The study area (Kirkuk) city is located in the north eastern part of Iraq at cross of longitude $(35^{0}2800")$ and Latitude $(44^{0}24'00")$. This study was conducted on drinking water that is used in this city, the samples were taken from Kirkuk unified water supply project include raw water (Lower Zab River), treated water (in side station)(WTP). the samples is also taken from water tanks type (GST2) (five main tanks supplied water to the city represents by five zones), and also from tap water of some house in (Zone3) as shown Figure 1.

This project was designed with a production capacity about (75,000) m^3 / h which contain from the following units:[20]

A- (Low lift pumping station): This station is located on the Kirkuk irrigation canal and is far about 1 km from tap station the water is pulled from Irrigation canal that contains from the below units:

1- Suction well 2- Low lift pump

B- Water Treatment Plant: which contain of (**1-Receiving well 2- Flash mixtures**) **3-**(**Clarifloculators**): (16) circular basins every basin have (42m) diameter and (9m) depth each basin contain two central basin internal basin for slow mixture(Sinter basin) external (sediment basin) that's water poured from this basin into a collection basin where chlorine is added as an initial stage before entering to filters.

4-(Filters): It is a type of quick (sand filters) that's work by gravity there where (80) filters that's water enter to it by (20) line of pipes every pipe connect to (4) filters, the capacity for each filter is about (5) $m^3/m^2/h$ and the area is (40) m^2 . Chlorine was added to filtered water and water was discharged from each filter in about (200) m^3/h our.

5-(Ground tanks): two ground tanks the capacity for each one is (16000) m^3 this tanks collect the water that's out from (Sterilization tank) before pumping water to the city by (high lift pumps)

6-(high lift pumping station): this station contains (7) pumps capacity for each one is $(2520) \text{ m}^3$ /hour, pressure load (40-80) m, this pumps connect by two cities, of pipes diameter for each pipe is (1200) mm through this station water pumped to the city tanks type (GST).



Figure 1-Location of Study Area

Methodology

4 water samples were collected from the study area included(Lower Zab river, treated water (WTP), the water of tanks(GST2), tap water (Zone3) in two seasons and stored in bottles type (polyethylene).pH, electrical conductivity (E.C), were measured in Kirkuk University environment research unit Laboratory. The samples filtered through pre-washed(0.45 μ m) nitrocellulose filters and acidify by using concentrated nitric acid to keep pH<2 and storied at-4⁰ to preserve the samples for accurate analysis of heavy metals [21]. The range of (pH) and (E.C) values are (7.5-8.5) and (250-300) μ s/cm for two seasons (Wet-Dry) 2016-2017, And these values are within the allowed limits by WHO (6.5-8.5) for pH, while for conductivity the organization did not specify the range [22][23]. The heavy metals concentrations were measured before and after treatment by using (Inductively Coupled Plasma Mass Spectrometry) technique in the (Bureau Veritas Mineral Laboratories – Vancouver Canada).

Results and discussion

The results of the current study show differences in the concentrations of selected heavy metals after treatment with chemicals and other mechanisms on which the station operates, the concentration of selected metals has become less than the concentration in raw water (before treatment) and this concentrations will not effect on human health because the results did not exceed the concentrations that permitted by WHO, just zinc exceed the allowed limits as shown in Table-1.

Metals	Lower za	ıb river	Treated water		WHO (2017)	USEPA(2017)
Heavy metals	Wet	Dry	Wet	Dry		
As	1.3	1.5	0.5	0.6	10	10
Cd	5.3	6.5	0.6	0.8	3	5
Cu	10.47	10.58	6.45	6.76	2000	1300
Fe	59.1	72	38	46	<3000	
Mn	6.3	7.7	4.6	5.5	400	
Ni	5.1	6.5	2.5	3.5	70	
Zn	21403.7	23916	6537	6675	3000	
Cr	1.2	1.5	0.58	1	50	100
Pb	3.6	4.2	1.4	2.4	10	15

Table 1-Concentration of heavy metals before and after treatment in (ppb)

in the Table-2 the results show the simple difference in the concentration of a specific element from one place to another within the area of interest (after treatment) in ppb included water of tanks type (GST) and tap water.

Metals	Water of ta	anks (GST)	(Tap water)		WHO (2017)	USEPA(2017)_
Heavy metals	Wet	Dry	Wet	Dry		
As	0.5	0.60	0.53	0.60	10	10
Cd	0.63	0.7	0.65	0.71	30	5
Cu	6.46	6.78	7.00	7.10	2000	1300
Fe	52.3	63.7	60.2	67.6	<3000	
Mn	4.4	6.7	4.4	6.8	400	
Ni	3.6	3.7	3.65	3.75	70	
Zn	6550	6680	7200	7320	3000	
Cr	0.6	1.1	0.8	0.9	50	100
Pb	2.5	2.6	2.7	2.75	10	15

Table 2-concentration of heavy metals in water of tanks, tap water in (ppb)

Health Risk Assessment

Risk assessment is a process which estimates the real or prospect reverse impact of contaminants to plants and animals and which condense on the disadvantage that has been or will be carried out by contaminants [24]. depending on the concentration of metals (non-carcinogenic risk) were assessed according to the exposure pathways of the previously recommended contaminants from(US environmental protection Agency). The potential water exposure pathways are direct water ingesting, and skin absorption of contaminants present in water. For calculation, the following equations were used [25].

Where (ADD= Average Daily Dose by ingestion and dermal absorption (ADD derm), unit in $\mu g/kg/day$), (BW= Average Body Weight (kg, 70 for adult and 15for children), (EF=Exposure Frequency(days/year, 350),(ED=Exposure Duration(years,30 for adult and 6 for children),(IR=Ingestion Rate(I/day, 2.2 for adults and 1.8 for children), Cw=concentration of the estimated metal in water ($\mu g/L$), (SA=exposed Skin Area(cm2, 18000 for adult and 6600 for children),(KP=dermal permeability coefficient in water cm/h) (ET=Exposure Time(h/day, 0.85 for adult and 1 for children), (AT=Averaging Time(day, for non-carcinogenic ED*365 for carcinogenic 70*365 for adult and children), (CF=unit Conversion Factor(L/cm3, 0.001).

Before a risk is described, hazard quotient (HQ) of each metal is calculated by using the following equation [26][27].

$$(HQ = ADD/RFD)$$
(3)

Where (HQ) is hazard Quotient via dermal or ingestion (unit less), RfD is Reference Dose via dermal or Ingestion in (μ g/kg/day) (non-carcinogenic risk) if; HQ>1 non-carcinogenic effects (concern).HQ< 1 accept level (no concern), (HI) value is Σ HQ _{ing}+ HQ _{derm}

As for carcinogenic influence, Cancer risk was calculated by using the following equation [28]:

(CR=ADD_{ing} * SF), (CR=ADD_{derm} * SF)

(4)

(SF= oral carcinogenic slope of the contaminant (mg/kg/day) for Pb=0.0085, for Cr=0.05[25]. If (CR) bigger than the range 10^{-4} - 10^{-6} carcinogenic influence (concern) but if (CR) smaller than 10^{-4} - 10^{-6} agreeable level (non-concern).[29]

(HQ)_{ing} (HQ) _{derm}, (HI) where calculated for water samples show in Tables-(1, 2) for two seasons (HQ) ing, (HQ) derm, (HI) values for, Cd, Cr, Pb, Cu, Mn, Fe, As, Ni is less than (1) for adult and children. this is indicates the effect of water treatment that's induced the concentration of heavy metals in water also meaning there is no health risks (non-carcinogenic) (non-concern), as show in Figures-(2-9) and for dry season is higher than wet season this is a normal result due to water level decline in summer because of the lack of rainfall, which leads to deposition of elements deep in the water supply[29]. but (HQ) ing, (HQ)derm, (HI) for Zinc metal(Zn) is higher than one before and after treatment this is may return to many factor like zinc known as one of the most element found in nature and there are many sources that's play to found this metal in high level like it's used in fungicides industries , detergents industries, drug industries because it has anti-bacterial properties which help to relieve stomach and bowel infections .[30][31] The study area is located within agricultural land where pesticides are used in large quantities, sewage water, and there are an oil production company within the same area.



Figure 3-(HQ)ing, (HQ)derm, (HI) values in drinking water for adult and children in Lower Zab river (Wet)

Figure 4-(HQ)ing, (HQ)derm, (HI) values in drinking water for adult and children in Lower Zab er (Dry)



Figure (4): (HQ)ing, (HQ)derm, (HI) values in drinking water for adult and children in torated water (wet)

Figure(5) (HQ) ing, (HQ) denn, (HI) values in drinking water for adult and children in treated water (Drv)



Figure(6) (HQ)ing, (HQ) derm, (HI) values in drinking water for adult and children in (water of tanks)((wet)

Figure(7) (HQ)ing, (HQ) derm, (HI) values in drinking water for adult and children in(water of tanks) (Dry)



Table-3 show carcinogenic risk(CR) for samples taken from Raw water (Lower Zab River), reated water) (water of tanks) type (GST) (Tap water) (CR) (CR) derm values for (Pb) and (Cr)

Table-3 show carcinogenic risk(CR) for samples taken from Raw water (Lower Zab River), (treated water), (water of tanks) type (GST),(Tap water), $(CR)_{ing}$ (CR)derm values for (Pb) and (Cr) in all samples for adult and children in two seasons not overtake the limits agreeable level (carcinogenic)(non- concern)

Water Type		Raw wat	ter (Wet)		Raw water (Dry)			
	CR	R ing CR d		lerm	CR ing		CR derm	
Heavy metals	Adult	Children	Adult	Children	Adult	Children	Adult	Children
Pb	3.59*10 ⁻⁷	3.02*10 ⁻⁷	1.09*10 ⁻⁸	4.43*10 ⁻⁹	4.43*10 ⁻⁹	3.52*10 ⁻⁷	1.28*10 ⁻⁸	5.16*10 ⁻⁹
Cr	$7.05*10^{-7}$	5.92*10-7	1.08*10-8	4.34*10-9	8.81*10 ⁻⁷	7.39*10 ⁻⁷	1.35*10-8	5.42*10-9

Table 3-A-Carcinogenic Hazards of heavy metals in Raw water (Lower zab river)

Water Type		Treated w	ater (Wet)		Treated water (Dry)			
	CR	ing	CR c	lerm	CR ing		CR derm	
Heavy metal	Adult	Children	Adult	Children	Adult	Children	Adult	Children
Pb	1.39*10 ⁻⁷	$1.17*10^{-7}$	4.28*10 ⁻⁹	1.72*10 ⁻⁹	1.72*10 ⁻⁹	2.01*10 ⁻⁷	7.33*10 ⁻⁸	2.95*10 ⁻⁹
Cr	3.41*10 ⁻⁷	2.86*10-7	5.21*10 ⁻⁹	2.10*10-9	5.87*10 ⁻⁷	4.93*10 ⁻⁷	8.98*10 ⁻⁹	3.62*10-9

Table 3-B-Carcinogenic Hazards of heavy metals in Treated water

Table 3-C-(Carcinogenic Hazards of heavy metals in (Water of tanks)

Water Type		Water of ta	anks (Wet)		Water of tanks (Dry)			
	CR	ing	CR o	lerm	CR	ing	CR derm	
Heavy metals	Adult	Children	Adult	Children	Adult	Children	Adult	Children
Pb	$2.49*10^{-6}$	$2.09*10^{-6}$	7.64*10 ⁻⁸	3.07*10 ⁻⁸	3.07*10 ⁻⁸	$2.18*10^{-7}$	7.94*10 ⁻⁹	3.19*10 ⁻⁹
Cr	3.5*10-7	$2.96*10^{-7}$	5.39*10 ⁻⁹	$2.17*10^{-9}$	6.46*10 ⁻⁷	3.87*10 ⁻⁷	$2.8*10^{-11}$	3.98*10 ⁻⁹

Table	3-B- C	arcinog	enic	Hazards	of heavy	metals i	n Tap	water
			,••	11000000				

Water		Tap wat	er (Wet)		Tap water (Dry)				
Туре									
	CR	R ing CR derm			CR ing		CR derm		
Heavy metal	Adult	Children	Adult	Children	Adult	Children	Adult	Children	
Pb	$2.69*10^{-7}$	$2.26*10^{-7}$	8.25*10 ⁻⁹	3.32*10 ⁻⁹	3.32*10 ⁻⁹	$2.31*10^{-7}$	8.39*10 ⁻⁹	3.38*10 ⁻⁹	
Cr	$4.7*10^{-7}$	3.95*10 ⁻⁷	7.19*10 ⁻⁹	2.90*10 ⁻⁹	5.28*10 ⁻⁷	$4.44*10^{-7}$	8.08*10 ⁻⁹	3.25*10 ⁻⁹	

Conclusion

After applied appropriate analysis by using modern technology it was found that drinking water in Kirkuk city was not contaminate with the following heavy metals (As, Cd, Cu, Fe, Mn, Ni, Pb, Cr), the concentrations that obtained after analysis of samples that taken from Kirkuk unified water supplied project as well as from the main tanks and also from tap water, was not exceeding the permissible limits according to WHO, the level of health risks of these metals were also assessed on human health and showed no significant effect. All concentrations obtained were less than (1) for (HQ)ing, (HQ) derm and (HI) for As, Cd, Cu, Fe, Mn, Ni, Pb,Cr.but (HQ)ing, (HQ) derm, and (HI) for Zn is higher than(1) and the little change in concentration for (Zn) return to the precipitations processes that's used in (WTP) but is not enough so we need to use aspecial units in (WTP) for heavy metals treatment. The carcinogenic hazards show no health risks for selected metals (Pb) and (Cr) **Recommendations**

1- Installation special and periodic monitoring units within (WTP) to measure the concentration of heavy metals with electronic save for outputs.

2- Connect a warning system on (WTP) shut down when the concentration of heavy metals exceeded the permissible limits.

3- Using nanomaterial's to treat drinking water because the big enormous potentiality for this technology to remove the pollutants from water[32,33]

4- Installation treatment units in townships that's flow the sewage water in to river directly without any treatment .

5- Replace the old and corroded pipes that's use to distribute water because pipes in general made of mixed of metals especially lead which consider one of the most threatening metal for our daily lives.[34,35,36]

6- Using landscape plants to reduce or to remove the heavy metals depend on many studies conducted by researchers[37,38,39]

7- Studying the other heavy metals which has been not addressed in this research

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