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A Study of some the optical properties for (SnO₂) thin films prepared by Sol-Gel method

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Abstract

In this research a bilateral tin oxide (SnO₂) thin films was prepared by Sol-Gel method, flow coating technique which is considered a simple and cheap technique. The horizontal sample preparation with and without the use of installed, and also with installed to vertical sample. Results of optical spectroscopy that the biggest values of the transmittance T was for horizontal films with an installed, as for the absorbance A was the biggest values of the vertical sample. For the reflectivity R was value in the horizontal films with installed, less than a horizontal films without Sticky as well as for films vertical. The results of the energy gap Eg for that direct allowed and forbidden transitions and for the indirect allowed and forbidden transfers for the studied Samples were also calculated, and they showed that the highest value of Eg was (3.85eV) for the direct forbidden transfers in the horizontal sample with installer, while the lowest value of Eg was (3.65eV) for the indirect allowed transfers without installer. The vertical film with indirect showed the lowest value for the energy gap ever (3.45eV) for the indirect allowed transitions.

Keywords: optical properties, Sol-Gel method

دراسة بعض الخواص البصرية لغشاء SnO₂ المحضر بطريقة التغطية بالسكب

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الخلاصة

تم في هذا البحث دراسة تحضير أغشية رقيقة ثنائي اوكسيد القصدير SnO₂ بتقنية المحلول الجيلاتيني (Sol-Gel) بطريقة التغطية بالسكب (Flow Coating) حيث تمتاز هذه التقنية بالبساطة وقلة التكلفة. تم تحضير عينة أفقية مع وبدون استخدام مثبت، وأيضاً مع مثبت لعينة عمودية. أظهرت نتائج المطيافية الضوئية أن أكبر قيم للنفاذية T كانت للأغشية الأفقية التي أضفنا لها مثبت، أما بالنسبة إلى الامتصاصية A فكانت أكبر قيم لها للعينة العمودية، وبالنسبة إلى الانعكاسية R فقد كانت قيمها أقل في الأغشية الأفقية مع مثبت، منها في الأغشية الأفقية بدون مثبت وكذلك لأغشية العمودية. كما بينت النتائج والحسابات في قيم فجوة الطاقة Eg في الانتقالات الالكترونية المباشرة المسموحة والممنوعة وفي الانتقالات غيرالمباشرة المسموحة والممنوعة للمدرسة، أن أكبر قيمة فجوة الطاقة للعينة الأفقية مع مثبت هي (3.88eV) وذلك في الانتقالات المباشرة الممنوعة، أما أصغر قيمة عرض فجوة الطاقة للعينة الأفقية بدون مثبت فهي (3.82eV)

وذلك في الانتقالات غيرالمباشرة المسموحة. بينما أبدت العينة العمودية أصغر قيمة لفجوة لطاقة على الاطلاق حيث بلغت (3.38eV) وذلك في الانتقالات غيرالمباشرة المسموحة .

Introduction

The increased importance of this technology because of It is characterized by low its cost, low loss material, suitability Broad surfaces [1].

This method is used for Cover glass equipment for protection, but can be used on flat glass. This method works with any type of substrates and characterized Films prepared in this way as the most thickness as compared with other methods.

Where they were used this method for the preparation of many types of thin films, such as zinc oxide (ZnO) copper oxide(CuO) tungsten oxide(WO₃)etc.. Tin oxides films possess a wide use, especially in the field of solar cells, Calendar vector transparent and gas sensors [9 causes of high permeability and good conductivity in these films to the chemical composition and structure to the microwave [2].

This research aims to determine the best conditions for the preparation of thin-film deposition of tin oxide by Sol-Gel method, flow coating technique, and study the transmittance, absorption and energy gap .Tin oxide status of horizontal and vertical and the effect of the installer DEA on these values, The importance of thin films oxide in the growing applications, where Used in solar cells, Photo Detectors and Transistors [3-5].

Experimental part

Dual- tin oxide with a white color material, molecular weight ((150.69) gm / mol) density ((6.95) gm / cm³) melting point ((1630 °C)) and boiling points (1800 - 1900) °C [1, 2].

Installation of dual- tin oxide are crystalline quartet installed and the lattice parameters are crystal size of the level (101) of the BRAC law was (d = 3.350Å) and the lattice constant (a = 4.736Å) [6-8].

A sol - gel, flow coating technique was used to prepared for prepare thin films of tin oxide deposited on the glass substrate, as it becomes tin chloride as a prefix , and the use of various organic solvent which is methanol, ethanol, propanol.

Abnormalities may occur in films where the substrates cleaned using ultrasound where we wash with a solution of a water and soap, then put it in the machine.

Ultrasound for a quarter of an hour, and then dried and make sure to get rid of the dust on the dry substrate.

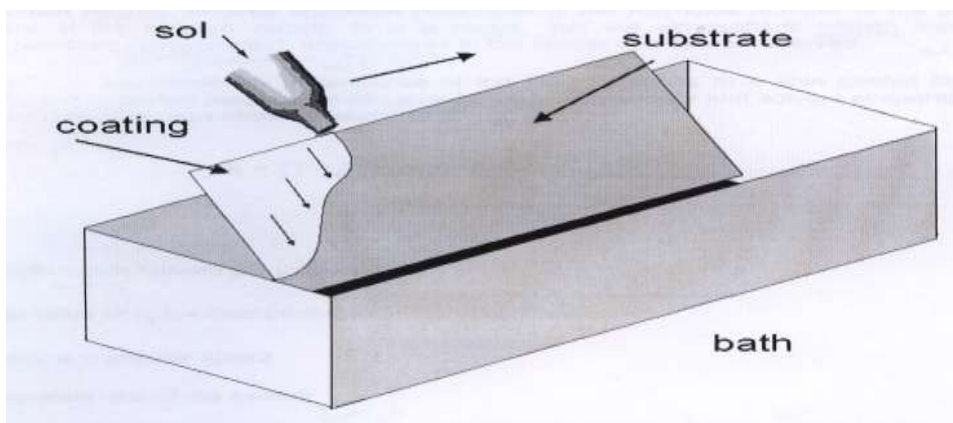


Figure 1- Flow Coating process

We then take three volumes of quad- tin chloride, And adding each one individually on each of ethanol , methanol , propanol then placed on a magnetic stirrer for one hour, we noticed This was the best melt with ethanol, so the dependence on ethanol in the preparation of the study samples.

Taken 5 g from the quad- tin chloride as a prefix, and resolves 20 ml and placed on the engine Magnetic for an hour, then left in the laboratory for a full day, and increase the accuracy of the

solution is filtered through a filter paper, and This becomes ready to use the flow of coating technology is divided into two parts with installer and without installer. And Samples set horizontal and vertical in three ways.

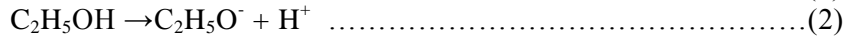
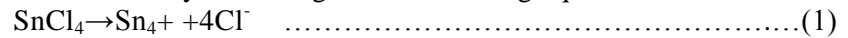
We poured the solution without the addition of installed on the glass slide placed diagonally around 5 degrees for detective table, and leave for 21 minutes and then put them horizontally on a table in the first sample.

Then pour the solution with the addition of installed on the glass slide placed diagonally around 5 degrees for laboratory table, and leaves for 21 minutes and then put them horizontally on a table, second sample.

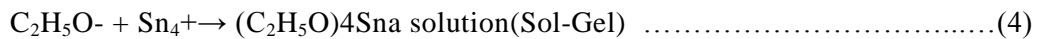
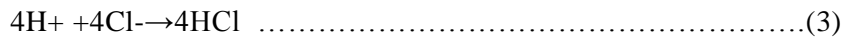
And in third sample we pour the solution with the addition of installed on the glass slide , which tend angle of 5 degrees For plumb for 21 minutes , and after slopping keep slide vertically with the laboratory table.

After the slides are placed in the electric oven at 400 °C and leave for an hour, and then pass on the slide to a dryer temperature of 200 °C for an hour, then placed in an atmosphere Detective conducted by optical study.

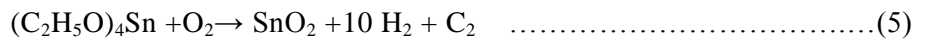
It can be described what happened chemically according to the following equations:



Then reacts



Then Heated to 400 °C



Reset and Dispassion

1- Structure measurements

In order to identify the crystalline structure of thin films SnO₂ X-ray diffraction technique (XRD) was used with source of (Cu. K) and a wavelength (0.015405Å).

It calculated the distance between the interplaner distains (d) using (Bragge Law) from the relationship: [10]

$$n\lambda = 2d \sin \theta \dots\dots\dots(6)$$

n : constant (rank diffraction) .

d : The interplaner distance.

λ: Wavelength of the incident beam .

θ :Diffraction angle with range about (20 ° -60 °) .

Measurements of the optical properties of which include absorbance, transmittance, reflective and the energy gap for SnO₂ , this follows deposite on glass substrate using a spectrometer the UV- visible (UV / VIS - PV- 8800 spectroscopy photometer) processor from a company (Pye Unicom)and took the transmittance of the spectroscope, The absorbance was calculated from [11].

$$A = \log(1/T) \dots\dots\dots(7)$$

the reflectivity is given by[12]

$$R = 1 - (A + T) \dots\dots\dots(8)$$

Then the absorption coefficient is given by

$$\alpha = 2.303A/t \dots\dots\dots(9)$$

where (t) is the thickness of the membrane [12,13]

The subside coefficient is given by

$$K_o = \alpha\lambda/4\pi \dots\dots\dots(10)$$

where (λ) is the length of the photon wave incident. [14]

the Refractive index was determined of the equation [15,16]

$$n_o = \sqrt{\left(\left(\frac{1+R}{1-R}\right)^2 - (K_o^2 + 1)\right) + \left(\frac{1+R}{1-R}\right)} \dots\dots\dots(11)$$

Figures- 2, 3 and 4 show the XRD patterns for SnO₂ thin film of horizontal and vertical Samples with and without installer. It is show that the films was Polycrystalline and the effect of installer was no upped that is agree with previous studies[1-4] ,There was no apparent effect for installer on the structure of the films. These results are consider with the (ASTM) (American Standard of Testing Materials [3- 6].

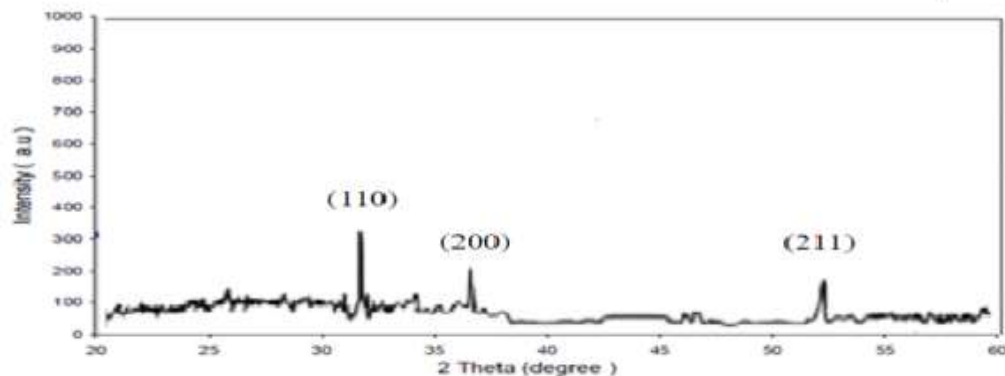


Figure 2- X-ray diffraction of (SnO₂) thin film of the horizontal sample without installer

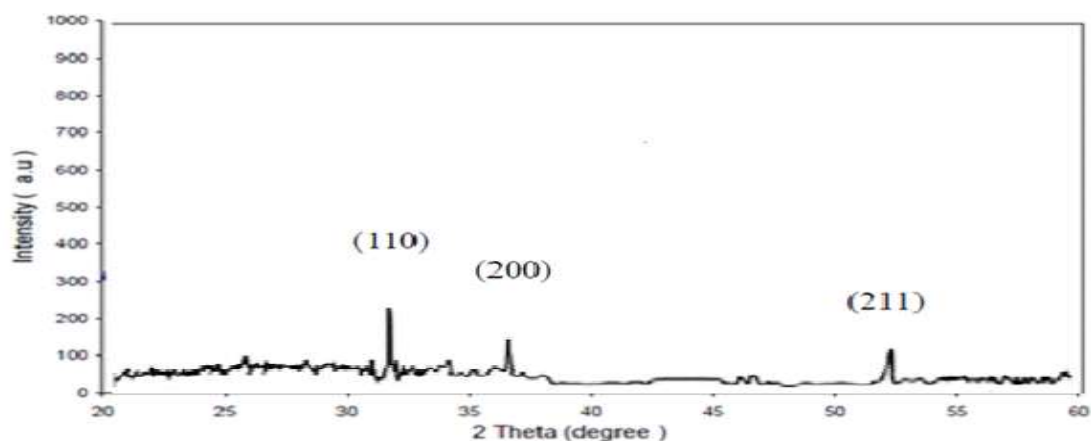


Figure 3- X-ray diffraction of (SnO₂) thin film of the horizontal sample with installer

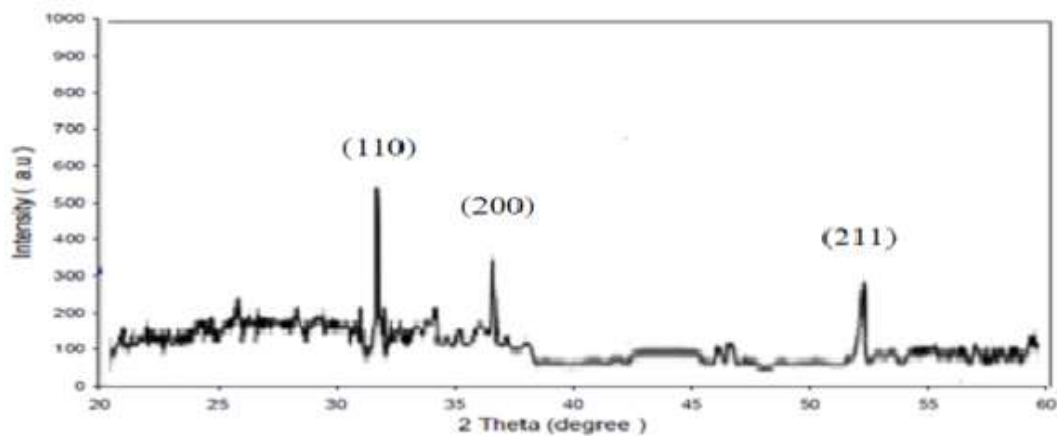


Figure 4-X-ray diffraction of (SnO₂) thin film of the vertical sample with installer

2-Optical Measurements

Figure- 5 and 6 show the Transmittance and Absorbance spectrum measured for SnO₂ thin films respectively. It is found the Transmittance increased with increasing wavelength. Between the phases (300-350nm) and reached to a high transactions values even greater than 90% toward the field of infrared.

It is found that the transmittance increased with increasing wave length in the rang (300-350) nm in all of samples but in the vertical samples it was smaller than horizontal samples of each wave length, as well as permeability values was the largest in horizontal sample which has been added the installer of the sample without installer, especially in the field between the phases (300-350) nm and the transmittance increased to a high values reached greater than 90% toward the field of infrared.

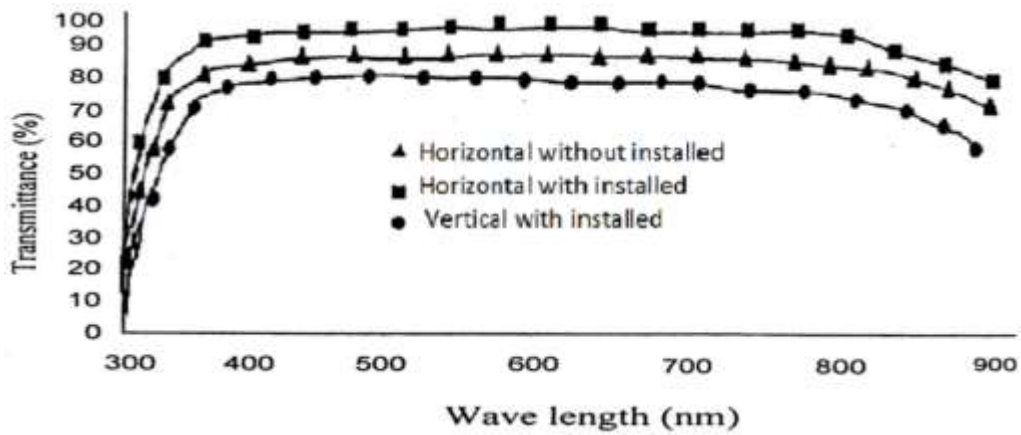


Figure 5- Transmission spectra of (SnO₂) thin film of horizontal and vertical Samples with and without installer.

Figure-6 represents the relation between absorbance and wave length of (SnO₂) thin film. In general it can see that the absorbance decreases with increasing of wavelength and the absorbance of the vertical sample is greater than horizontal samples , clearly the absorbance of horizontal samples that without installer is smaller than the sample of horizontal which with an installer.

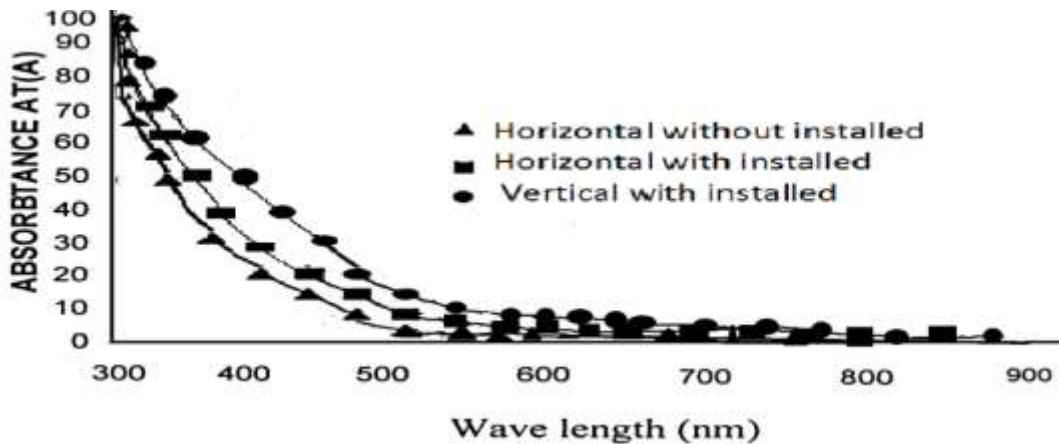


Figure 6-Absorbance spectra of (SnO₂) thin film of horizontal and vertical samples with and without installer.

Figure-7 shows that R as a function λ for SnO₂ , It is found that R decreases with increasing of wavelength in the rang (380-900) nm , and the value of R of the vertical samples is greater than

the horizontal samples as in Figure-7 ,But the summit has withdrawn from 300nm horizontal samples to 400nm vertical samples , where the reflectivity increase rapidly and abruptly and went to the value (20.3%) of the samples for horizontal and vertical , that all of the reflection and absorption was begun for horizontal sample at 300nm , while the vertical sample 400nm due crystalline structure according to the method deposition. As was the reflectivity values for horizontal sample with installed smaller than the horizontal sample without installed. When wavelength 600nm transmittance 79.81% vertical sample and absorbance 9.53% and reflectivity 10.32 % for the same sample which fulfilled the basic relationship $R = 1 - (A + T)$ At this wavelength the same 600nm horizontal sample with a fixed value of transmittance 95.16% and absorbance 2.12 % and reflectivity 2.63 % so also realized previous relationship. These values for each of the transmittance absorbance and reflectivity converge with the values contained in the previous global scientific studies.

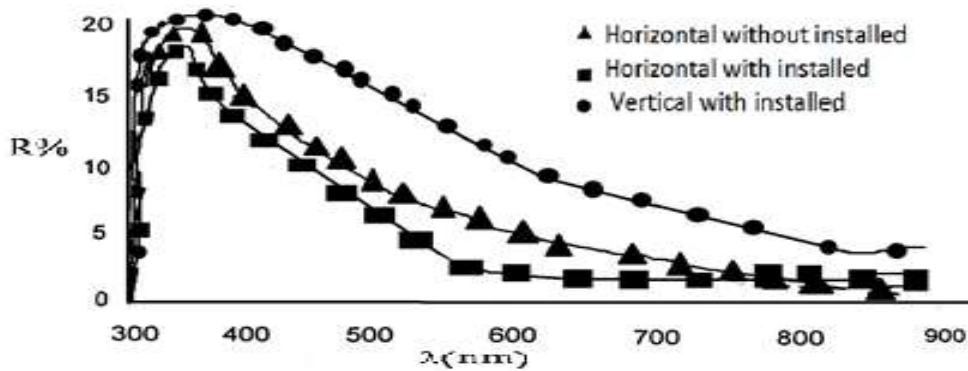


Figure 7- reflectivity spectra of (SnO₂) thin film of horizontal and vertical Samples with and without installer.

Figure- 8 represents the relation between K_0 and $h\nu$ of (SnO₂) thin film. It is note from Figure-8 that the coefficient values which is found from relation(10) increase with increasing of photon energy, and that the value of coefficient of damping in the vertical sample is greater than the horizontal samples at the same energy values always, due to a difference form the crystalline structure between the horizontal and vertical modes. We also find that the damping coefficient values are almost identical in the two Horizontal samples, and this shows that the installer did not have an active role enough in damping coefficient values.

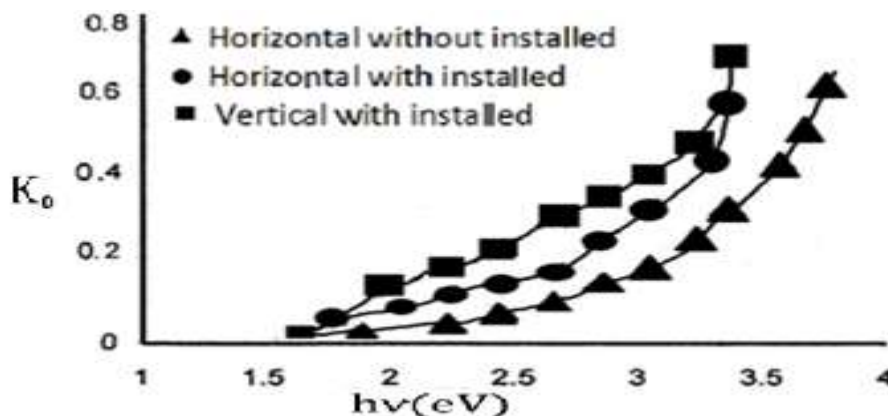


Figure 8-the absorption coefficient of (SnO₂) thin film of horizontal and vertical Samples with and without installer

Figure -9 represents the relation between n_o and $h\nu$ of (SnO_2) thin film, This Figure shows that n_o as a function of $h\nu$ that the refractive index values of vertical sample is greater than the horizontal values for samples, The values ranging between (1.53–2.17) which is the value of the refractive index of the tin oxide film moderation (1.85), and this is in line with the values, Calculated in the previous global scientific agenda [16], which was (1.80).

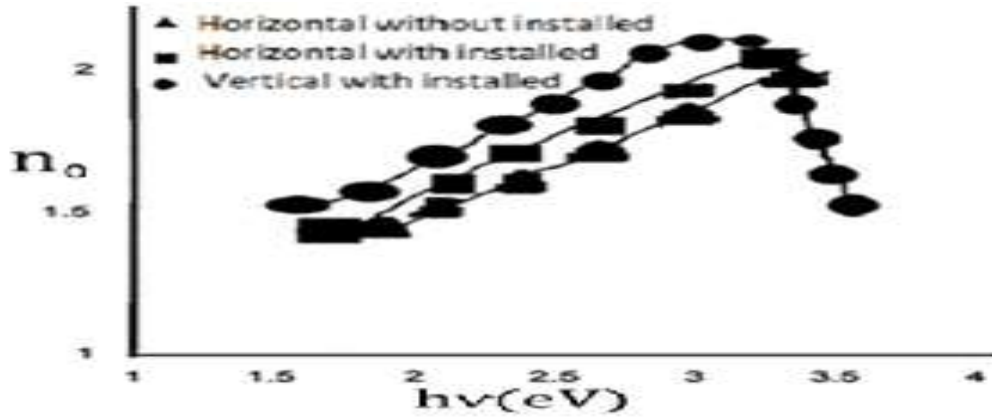


Figure 9- the refractive index of (SnO_2) thin film of horizontal and vertical Samples with and without installer.

Figures- 10,11and 12 shows that real dielectric constant and the imaginary dielectric constant as a function of $h\nu$, In the horizontal and vertical position there are similarity of the behavior of the refractive index with a real dielectric constant as for the section of the imaginary dielectric constant , the difference in the curves between the horizontal and vertical modes due The difference in absorption coefficient values , The imaginary dielectric constant is a measure of lost Radiation energy incident on the sample , It represents a contribution of charge carriers within the stated article in the absorption of the optical radiation energy. [17-19]

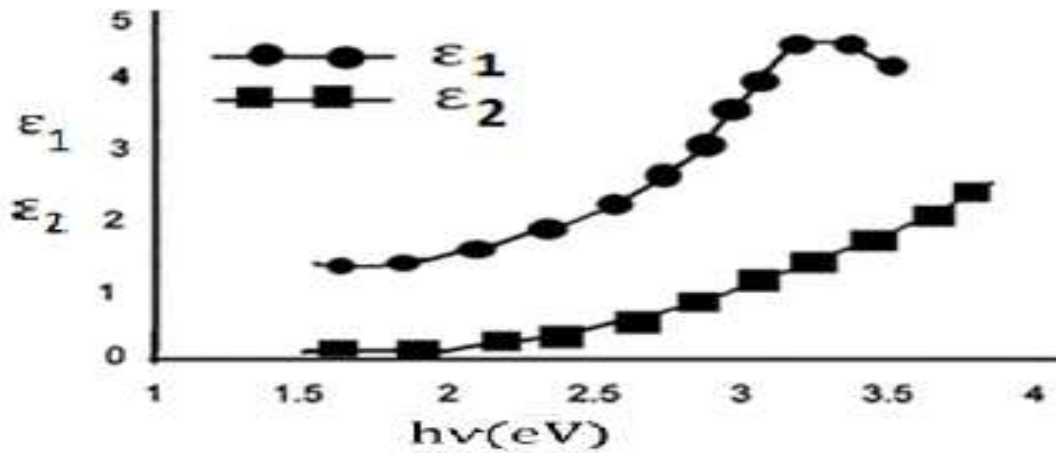


Figure 10-the real dielectric constant and the imaginary dielectric constant of (SnO_2) thin film of the horizontal sample without installer.

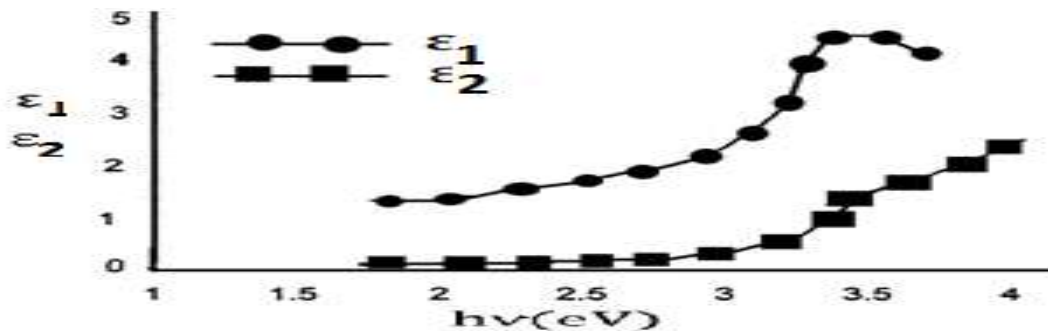


Figure 11- the real dielectric constant and the imaginary dielectric constant of (SnO₂) thin film of the horizontal sample with installer

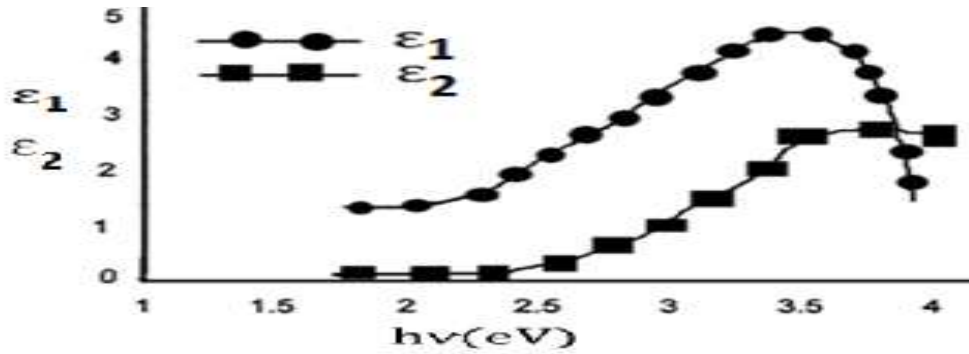
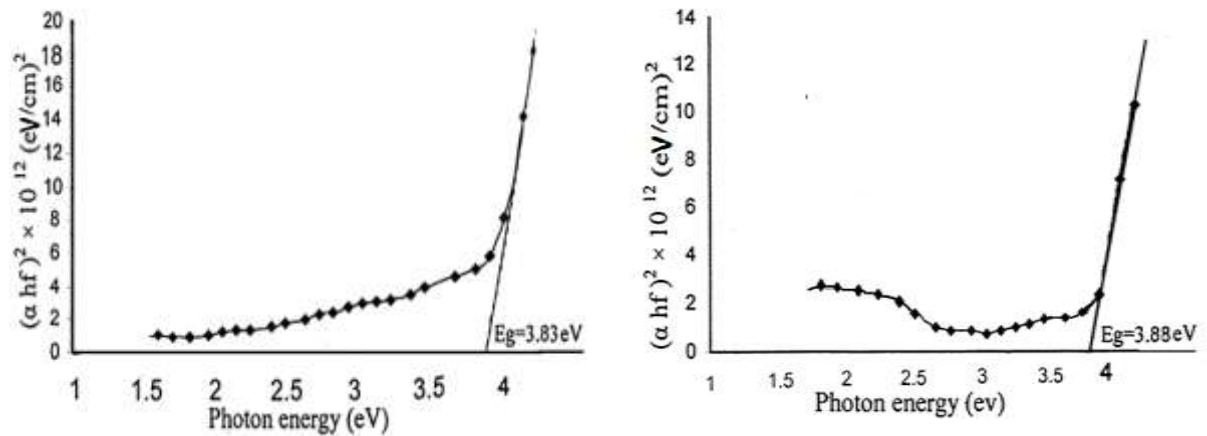


Figure 12- the real dielectric constant and the imaginary dielectric constant of (SnO₂) thin film of the vertical sample with installer.

Figures-13-18 show the direct and indirect transfer of both types of allowed and, forbidden and the energy gap which found from the figure conclude the following:

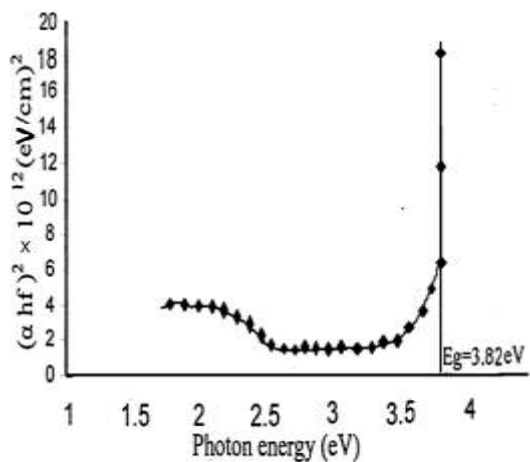
For vertical sample energy gap decreased for each transfer compared to horizontal samples, and arrived at (3.38eV) in the transfer of indirect allowed either in horizontal samples were smaller value in the transfer of indirect allowed and which were (3.68eV), The largest value gap energy of all samples (3.88eV), The horizontal sample with installer and these results are comparable with research [20,21].



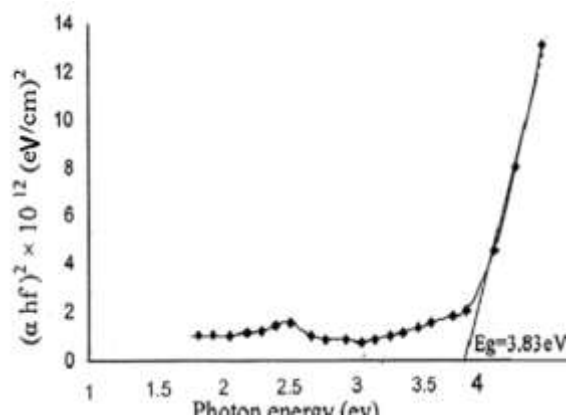
a- allowed

b- is forbidden

Figure 13- the energy gap of (SnO₂) thin film of the horizontal sample with installer of direct transitions a- allowed b- is forbidden.

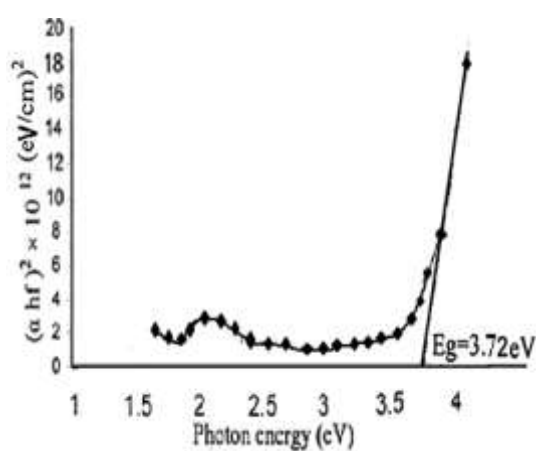


a- allowed

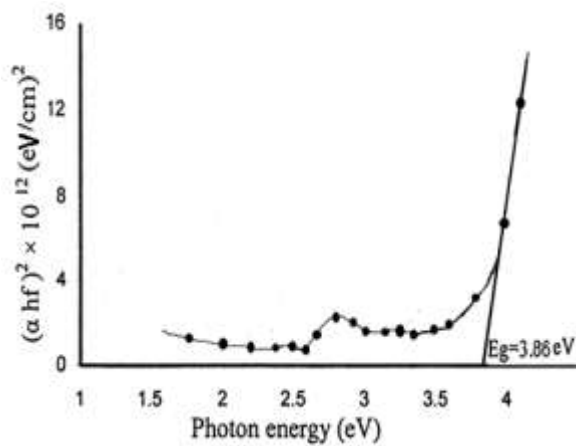


b- is forbidden

Figure 14 - The energy gap of (SnO₂) thin film of the horizontal sample without installer of direct transitions a- allowed b- is forbidden.

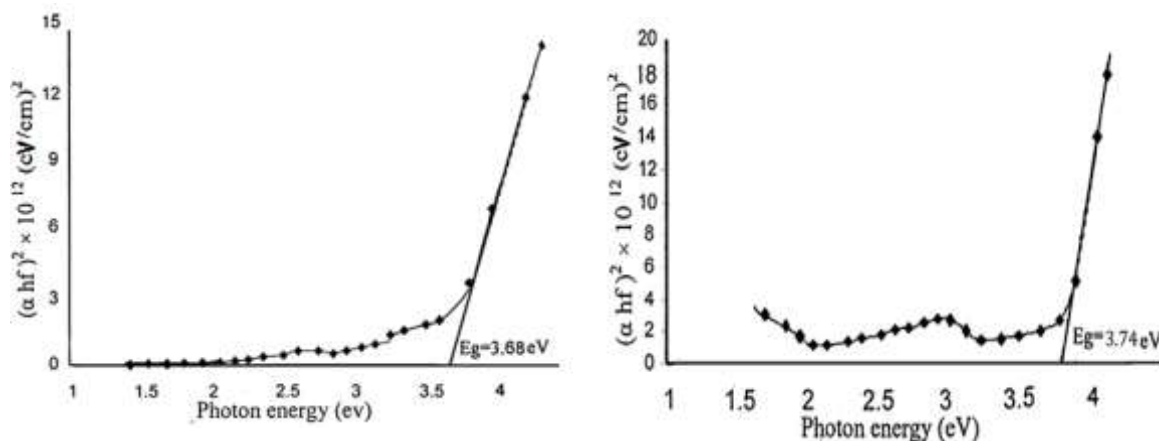


a- allowed



b- is forbidden

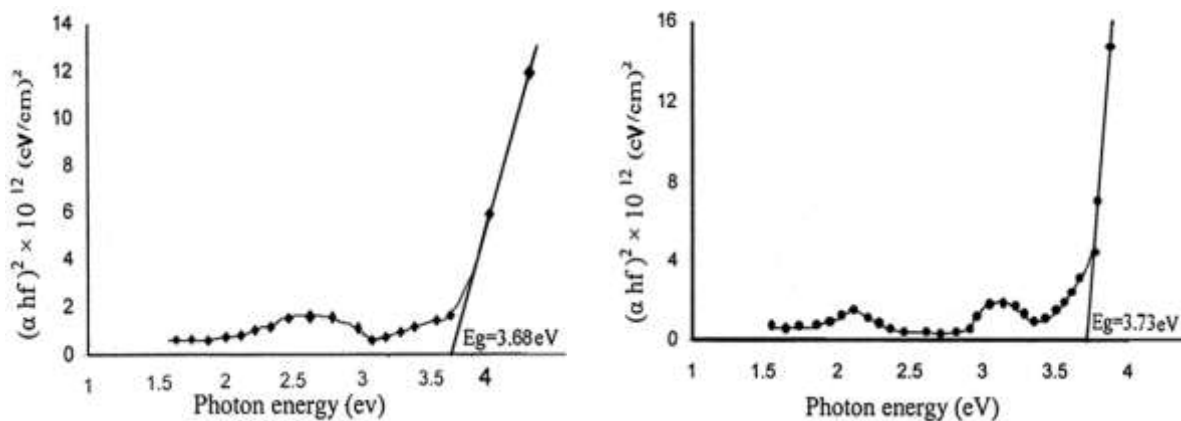
Figure 15- the energy gap of (SnO₂) thin film of the vertical sample with installer of direct transitions a- allowed b- is forbidden



a- allowed

b- is forbidden

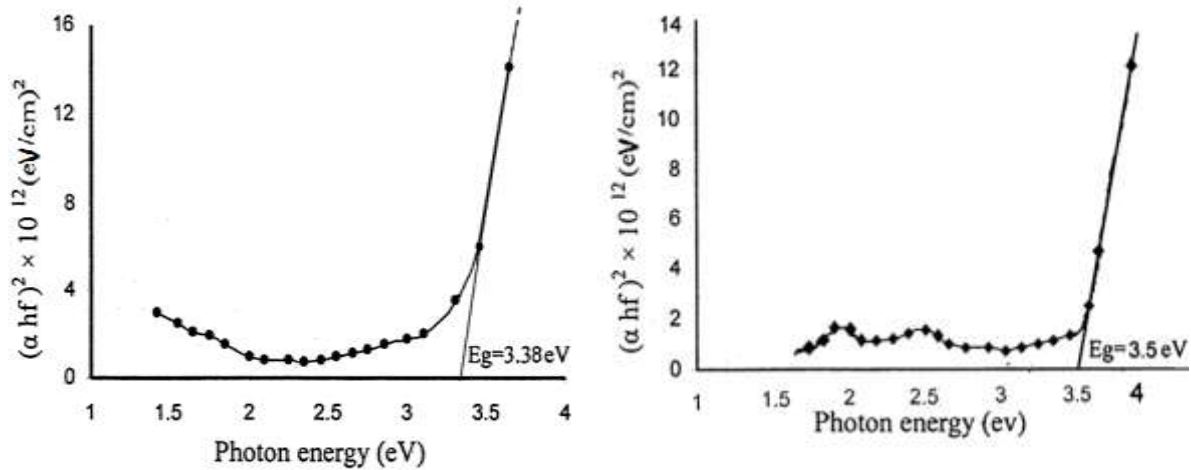
Figure 16- the energy gap of (SnO₂) thin film of the horizontal sample with installer of indirect transitions a- allowed b- is forbidden.



a- allowed

b- is forbidden

Figure 17- the energy gap of (SnO₂) thin film of the horizontal sample without installer of indirect transitions a- allowed b- is forbidden



a- allowed

b- is forbidden

Figure 18- the energy gap of (SnO₂) thin film of the vertical sample with installer of indirect transitions a- allowed b- is forbidden

Conclusions

We found that the high transmittance ranging from (80-95)% within the wavelength (600-900)nm, The absorbance ranging (10-20)% from the same field, means we have a film with a weak absorbency, The results indicate that the tin oxide film is high absorbance only in a narrow field is the field of UV rays.

Transmittance become better with installer of horizontal samples, and a few improve the absorbance of the sample vertical, vertical samples gave lower transmittance and upper Absorption coefficient compared with the horizontal samples.

The results show that the biggest value of the energy gap in the direct transitions allowed (3.88eV), while the smaller the value of the energy gap in the transfer of indirect allowed (3.82eV).

The vertical sample has a smaller value of the energy gap (3.38eV) in the transfer of indirect allowed. However, the vertical position gave lower values for the energy gap of the four transitions, thus the biggest values of the refractive index.

References

1. Dominguez, J. E. Pan, X. Q. and Fu, L. **2001**. Epitaxial SnO₂ thin Films Grown on (1012) sapphire by Femtosecond pulsed laser deposition. *J. Appl. Phy*, **2**(3): 2340-2346.
2. Elangovan, E. and Ramamurthi, K. **2003**. Studies on optical properties of polycrystalline SnO₂: Sb thin films prepared using SnCl₂ precursor. *Cryst. Res. Technol.*, **38**(9): 779-784.
3. Roy, S. S. and Podder, J. **2009**. Studies on Tin Oxide (SnO₂) And Cu Doped SnO₂ Thin Films Deposited by Spray Pyrolysis Technique for Window Materials in Solar Cells. International Conference on Mechanical Engineering.
4. Rezvani, H. **2010**. The effect of deposition parameters on the sensing behavior of the SnO₂: Cu nano-structure thin films including CO₂-gas sensor. *Indian Journal of Science*, **3**(6): 0974-6846.
5. Arivazhagan, V. and Rajesh, S. **2010**. Preparation of Nanocrystalline SnO₂ Thin Films For Micro Gas Sensors. *Journal of Ovonic Research*, **6**(5): 221-226.

6. Subramanian, N. Sankara, Santhi, B., Sundareswaran, S. and Venkatakrisnan, K. S. **2006**. Studies on Spray Deposited SnO₂, Pd:SnO₂ and F:SnO₂ Thin Films for Gas Sensor Applications. *Metal-Organic and Nano-Metal Chemistry*, **36**: 131-135.
7. Lee, S. Uk, Boo, J. and Hong, B. **2011**. Structural, Electrical, and Optical Properties of SnO₂:Sb Films Prepared on Flexible Substrate at Room temperature. *Japanese Journal of Applied Physics* **50**, Suwon, Korea, **4**: 440-746.
8. Park, J. C., Hwang, S., Kim, M. J., Kim, J. K., Lee, W. Y., Park, J. S., Kim, E. H., Jung, Y., Shim, K., Cho, H. **2009**. Anisotropic pattern transfer in SnO₂ thin films for the fabrication of nanostructure-based gas sensors. *J. Ceram Process Res.* **10**(6): 827-831.
9. Adnan, M. Munir, A. and Maher, G. **2013**. Study the structural properties ZnO of thin films of pure and tinged of Technology Sol-Gel. *Journal of Aleppo Research, Basic Science Series* **9**(6): 121 - 124.
10. Callister, W. D. **1997**. *Materials Science and Engineering*. 4th Edition, The United States of America.
11. Fadil K. Dahsh, Khudair A. Mushajji. **2011**. Study of the effect of Co- deflation on Tin Oxide film. *Journal Of College Of Education*, **17**(1): 131.
12. Hussian A. H., Shymaa K. **2013**. Structural and optical characterization of Nano crystalline SaO₂ Thin film prepared by spray pyrolysis technique. *Journal of kyfa physis*. **5**(1): 316-322.
13. Korotecenkov, G. Brinzari, V. **2002**. Material Sensors and Engineering. *Material Sensors and Engineering*, **C19**: p.73 .
14. Chopra, K. L. **1969**. *Thin Film phenomena. Journal of Physics, Condensed Matter*, M.C.Graw-Hill.
15. Ilıcan, S. Caglar, Y., Caglar, M. and Demirci, B. **2008**. Polycrystalline indium-doped ZnO thin films: preparation and characterization. *Journal of Optoelectronics and Advanced Materails*, **10**(10): 2592-2598.
16. Al Shammari, Amanuilah F. M. **2010**. Optical Characteristics of NiO Thin F: dm On glass formed by chemical spray pyrolysis. *Journal of kyfa , physis*, **2**(1).
17. Ravindran L. and Gómez, J. **2012**. Structural and optical characterization of Nano crystalline SaO₂ Thin film prepared by Sol-Gel technique. *Sol-Gel Sci. Technol.*, **61**(1): 2942-2953.
18. Mahmood, N. A. **2007**. A Study of the Structural and Optical Properties of SnS₂: Cu Thin Films Prepared by the Chemical Spray Pyrolysis. MSc. Thesis, University of Baghdad, College of Science for Women, Baghdad, Iraq.
19. Hamaguchi, C. **2001**. *Basic Semiconductor physics*. Book, 2nd Edition, Springer.
20. Basim E. Gaseous , Nawroz I. Hassan , Cheman B. Ismaeil, Hayfa G. Rashid. **2012**. Structural and Optical Constants of Fe-doped SnO₂ Thin Film. *Tishreen University Journal for Research and Scientific Studies*, **1**: 29-40.
21. Mishjil, K. A. **2005**. The effect of acceptor dopant concentration of (Al⁺³) on optical properties of (SnO₂) transparent conducting thin films. MSc. Thesis ,College of Education, Al-Mustansiriyah University, P.(158-177).