



ISSN: 0067-2904

Synthesis and Characterization of Gallium Oxide Nanoparticles using Pulsed Laser Deposition

Mohammed M. Hameed^{*1}, Abdul-Majeed E. Al-Samarai¹, Kadhim A. Aadim²

¹Department of Physics, College of Education for Pure Science, University of Tikrit, Iraq

²Department of Physics, College of Science, University of Baghdad, Iraq

Received: 23/10/2019

Accepted: 20/12/2019

Abstract

In this paper, the productions of gallium oxide (Ga_2O_3) nanoparticles were achieved via using the Nd: YAG laser deposition method with a fundamental wavelength (1064 nm). These nanoparticles were characterized by using different methods such as X-ray diffractometer (XRD), atomic force microscopy (AFM) and Ultraviolet-visible (UV-vis) spectroscopy. To examine the effects of laser energy on the properties of nanoparticles, the experimental results and theoretical considerations were prepared by the effective method of pulse laser deposition. The synthesis of Ga_2O_3 NPs was achieved with different ranges of energies (500 to 900 mJ). Average crystallite sizes of the synthesized nanoparticles were found to be in the range of 15-37 nm. On the other hand, the AFM images showed that the size is ranging between 60 and 85 nm. Optical parameters of the samples showed a strong dependence on average crystallite size.

Keywords: Gallium Oxide, Nanostructure, pulse laser deposition, Structural and Optical properties

تخليق وتوصيف الجسيمات النانوية لأوكسيد الغاليوم المحضرة باستخدام تقنية الترسيب بالليزر النبضي

محمد مجيد حميد^{*1}, عبد المجيد عيادة ابراهيم¹, كاظم عبد الواحد عادم²

¹قسم الفيزياء, كلية التربية للعلوم الصرفة, جامعة تكريت, العراق

²قسم الفيزياء, كلية العلوم, جامعة بغداد, العراق

الخلاصة

في هذا البحث, تم تحضير جسيمات اوكسيد الكاليوم النانوية باستخدام الترسيب بالليزر النديميوم ياك النبضي ذي الطول الموجي (1064 nm). وتم توصيف هذه الجزيئات بواسطة حيود الاشعة السينية, مجهر القوة الذرية, و التحليل الطيفي للأشعة فوق البنفسجية. لغرض فحص تأثير تغيير طاقات الليزر على خصائص هذه الجزيئات النانوية والتحقق منها, وباعتماد على الفرضيات النظرية اعدت الجزيئات النانوية باستخدام الترسيب بالليزر النبضي. وتم التحضير ضمن مدى من الطاقات يتراوح من 500-900 mJ. وكان معدل الحجم للجسيمات البلورية يتراوح بين (15-37 nm).

من جانب اخر, فان نتائج فحص النماذج بواسطة مجهر القوة الذرية بين ان حجم الجسيمات النانوية يتراوح بين (60-85 nm). كما اظهرت الخصائص البصرية اعتمادا كبيرا على الحجم البلوري

*Email: mohammedaz365@gmail.com

1. Introduction

Nanoscience is a wide area of science that studies and uses different techniques to obtain stable and new means of nanoparticles. In comparison with original and parent materials, nanoparticles become important and achieve more significance because of their remarkable physicochemical characteristic. Physical forces in the hold of nanoscale particles are included in the manual method of nanoparticles, which lead to the formation of large, stable and well-defined nanostructures. This method involves basic techniques such as vapor condensation, physical fragmentation, amorphous crystallization and many others [1]. There is an important role of metal oxide nanostructures and they have responsibility in several areas of chemistry, physics and materials science. Several applications are allowed and enabled by the different and significant properties of oxides in the production of circuits of microelectronic , devices of piezoelectric, fuel cells, sensors, coatings against corrosion and as catalysts (some catalysts have an active phase of oxide) [2]. Gallium oxide is considered as a wide band-gap semiconductor which has significant physical properties that are important for photocatalysts, gas sensors, ultraviolet photodetectors, and power devices [3]. These properties can be combined, which allows Ga₂O₃ thin films to be used in different applications, such as gas sensors, solar cells, deep-UV photo detectors, spintronics and field-effect transistors. Different techniques and methods can be used to prepare Ga₂O₃ films; for instance, electron beam evaporation, magnetron sputtering, molecular beam epitaxy, metal-organic chemical vapor deposition (MOCVD), sol-gel process, pulsed laser deposition and vapor phase epitaxy. In recent years, the huge potential of Ga₂O₃ power devices for future high-power and high-voltage applications have been illustrated by the important progress made in Ga₂O₃ transistors and diodes [4]. Numerous different polymorphs can be created from Ga₂O₃, which are designated as α -, β -, γ -, δ -, and ϵ - Ga₂O₃. β - Ga₂O₃ has a monoclinic structure and belongs to the space group C2/m [5].

The Ga₂O₃ material is a semiconductor material with a wide bandgap (4.3-5.0 eV) [3]. The main type of gallium oxide is the β -form. This is the most popular and well established gallium oxide polymorph. Through the whole temperature range till the melting point, β - Ga₂O₃ is the only stable polymorph, while the other polymorphs at temperatures above 750-900 °C can be transformed into the β - Ga₂O₃ because they are metastable [6]. It is possible to produce and create giant single crystals and films with epitaxia by high temperature processes, including melt crystallization and vapor epitaxy, because of the thermal stability of β - Ga₂O₃. Due to the availability and remarkable properties of β - Ga₂O₃, it has become more interesting among other polymorphs of Ga₂O₃ and many studies were conducted about its applications. In comparison with other Ga₂O₃ polymorphs, β - Ga₂O₃ is well characterized, while many variations still exist in published material property data [6,7]. The aim of this study is to investigate the effects of changing laser energies on the structural and physical properties of the film precipitation.

2. Materials and methods

In this project, gallium oxide powder was purchased from Sigma–Aldrich with a purity of 99.9%. The powder was pressed and transformed to pellets with 0.2 cm thickness and 1.5 cm diameter. Hydraulic piston (type SPECAC) was used for 15 min at a pressure of 6 tons. Glass substrates (2.5×7.5 cm) were used to deposit the films of gallium oxide. Distilled water was used to clean these glass substrates and remove the remaining dust and grime from their surface. Then, glass substrates were cleaned in alcohol for 15 min by an ultrasonic system to remove some oxides and grease. The air of plowing was used in this process to dry these glass substrates. Finally, soft paper was used to wipe the slides and Nd: YAG laser was applied to deposit films at room temperature utilizing pulse laser deposition (PLD) technique at a wavelength of $\lambda = 1064$ nm and energy from 500 to 900 mJ. The frequency of repetition was 6Hz for 200 shots which occurred on the target surface at an angle of 45°. The distance between laser and target is 10 cm, under vacuum and a pressure of 2.5×10^{-2} mbar. The most interesting method for deposition of metals and semiconductors is pulse laser deposition (PLD). DC magnetron sputtering provides the ability to rapidly deposit huge amounts of materials, delivering high levels of deposition and good coverage. Determination of the optimal location for the target range was achieved on the surface holder.

The evaluation of the structural, optical and electrical characteristics of the thin films was performed by using different characterization techniques. Measurement of optical interference fringes can be a quick, accurate, and non-destructive approach which can be used to determine the film thickness. This method uses the laser He-Ne of 632 nm, at 45° angle of incidence, to calculate the

thickness of Ga₂O₃ thin films. This approach depends on the laser beam interference reflected from the thin film surface and then substrates. The film thickness was determined using the following equation [8]:

$$t = \frac{\lambda \Delta x}{2 \chi} \quad \dots\dots\dots (1)$$

X-ray diffraction (XRD) is considered as one of the preferred and effective methods for this objective of the qualitative and quantitative evaluation of crystalline compounds. This experimental method has long been used to quantify and establish the overall structure of thin Ga₂O₃ films, including lattice constants, grain size recognition of unknown materials, single crystal orientation, and polycrystals orientation. In this research, X-ray diffraction XRD with Cu-K α X-ray tube ($\lambda = 1.54056$) was used to analyze thin films. The X-ray scans were performed between 2θ XRD values at a scanning frequency of $0.08333^\circ \text{ s}^{-1}$ with a diffraction angle of 2θ and a distance of $20^\circ - 80^\circ$.

Micrographs (Digital Instruments, CSPM-AA3000) of the Atomic Force Microscopy (AFM) were used to analyze the surface roughness and topography of the coated thin films.

UV companion SP-8001 double beam spectrophotometer, purchased from Metertech Corporation (Taipei, Taiwan) and covering a wavelength range of 190-1100 nm was employed. This system was used to prepare Ga₂O₃ thin films and achieve the optical measurements. There was a blank glass slide in one direction of the beam during the scanning, while another glass slide with the film deposit was put in the direction of the other beam. The films deposited on the glass slide substrates can lead to the spectrum of absorption displayed by the double beam spectrophotometer of the UV mate SP-8001.

3. Results and discussion

3.1. XRD) analysis of structural properties

Figure-1 below describes the XRD patterns of the PLD of Ga₂O₃ thin films. The diffraction peaks are indexed by comparing the data with JCPDS card files No.96-200-4988. Diffraction patterns revealed that thin films are strongly oriented along the (111) plane with favored orientation to the c-axis and monoclinic crystal structures. There were also low intensity peaks corresponding to planes (401) and (31-1⁻). Also, all intensities could be increased by increasing energy deposition. Bragg's law equation was used to estimate the spacing d of the lattice, as shown below [9]:

$$d = \frac{n\lambda}{2 \sin \theta} \quad \dots\dots\dots (2)$$

where n is the diffraction order which is taken as 1, λ is the X-ray wavelength used (0.154056 nm) for Cu K α target, and θ is Bragg peak diffraction angle (002). Lattice spacing d was calculated by the following relationship for the monoclinic lattice [10]:

$$\frac{1}{d^2} = \frac{1}{\sin^2 \beta} \left(\frac{h^2}{a^2} + \frac{k^2 \sin^2 \beta}{b^2} + \frac{l^2}{c^2} - \frac{2 hl \cos \theta \beta}{ac} \right) \quad \dots\dots\dots (3)$$

where h , k and l are the plane's Miller indices, while the lattice constant monoclinic unit cell was represented by a and c . The constants of the lattices a and c were determined using equation (3).

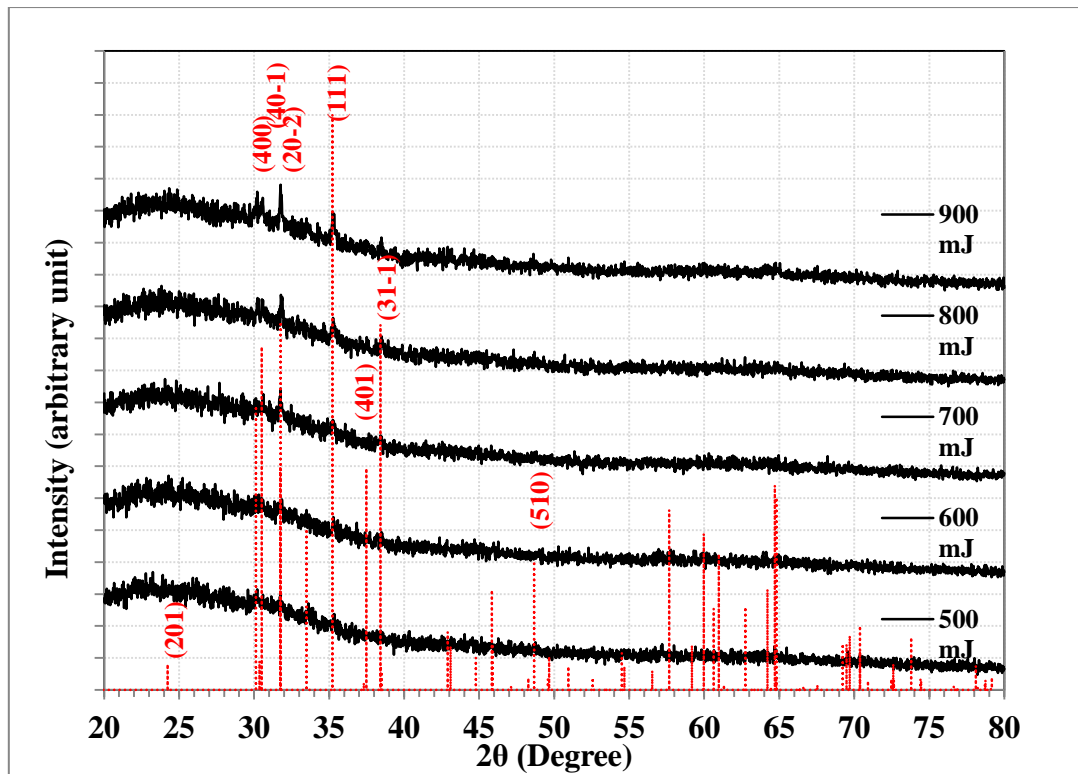


Figure 1-X-ray pattern of the PLD of Ga₂O₃ thin films.

The crystallites deficiency could attribute intrinsic stress during the growth. The intrinsic stress in the thin films can be caused by the deposition parameters such as growth temperature, gas pressure, laser power and pulse duration [9]. The negative sign with the stress value represents the inherent stress of the compression. The increasing in the thickness of film can lead to increase the stress. The energy of dislocation is inversely proportional with the film thickness, which means if the energy of dislocation is high the thickness of film is low. The increasing of the film thickness can lead to elevation of the density of dislocation and the stacking fault probabilities, thus the thin films stress was increased. Scherrer equation with XRD data was used to calculate the crystallite size (l) [11]:

$$l = \frac{0.9\lambda}{\beta \cos \theta} \dots\dots\dots (4)$$

where the full width at half maximum of the peak is represented by β. There was an improvement in the structural properties resulting from decreasing the size of the crystallite to increase thin film thickness.

3.2. Surface morphology and roughness

A photograph of Ga₂O₃ film was produced by atomic force microscopy (AFM) and showed a diameter of 2 inch, as shown in Figure- 2. The film had a strong granular surface consisting of an irregular grain agglomeration of tens of nanometers. Root mean squared roughness (R_{RMS}) for Ga₂O₃ thin films was 7.5 nm (Average of Three 1 mm 1 mm) near other observed values (3 to 8 nm) [12,13].

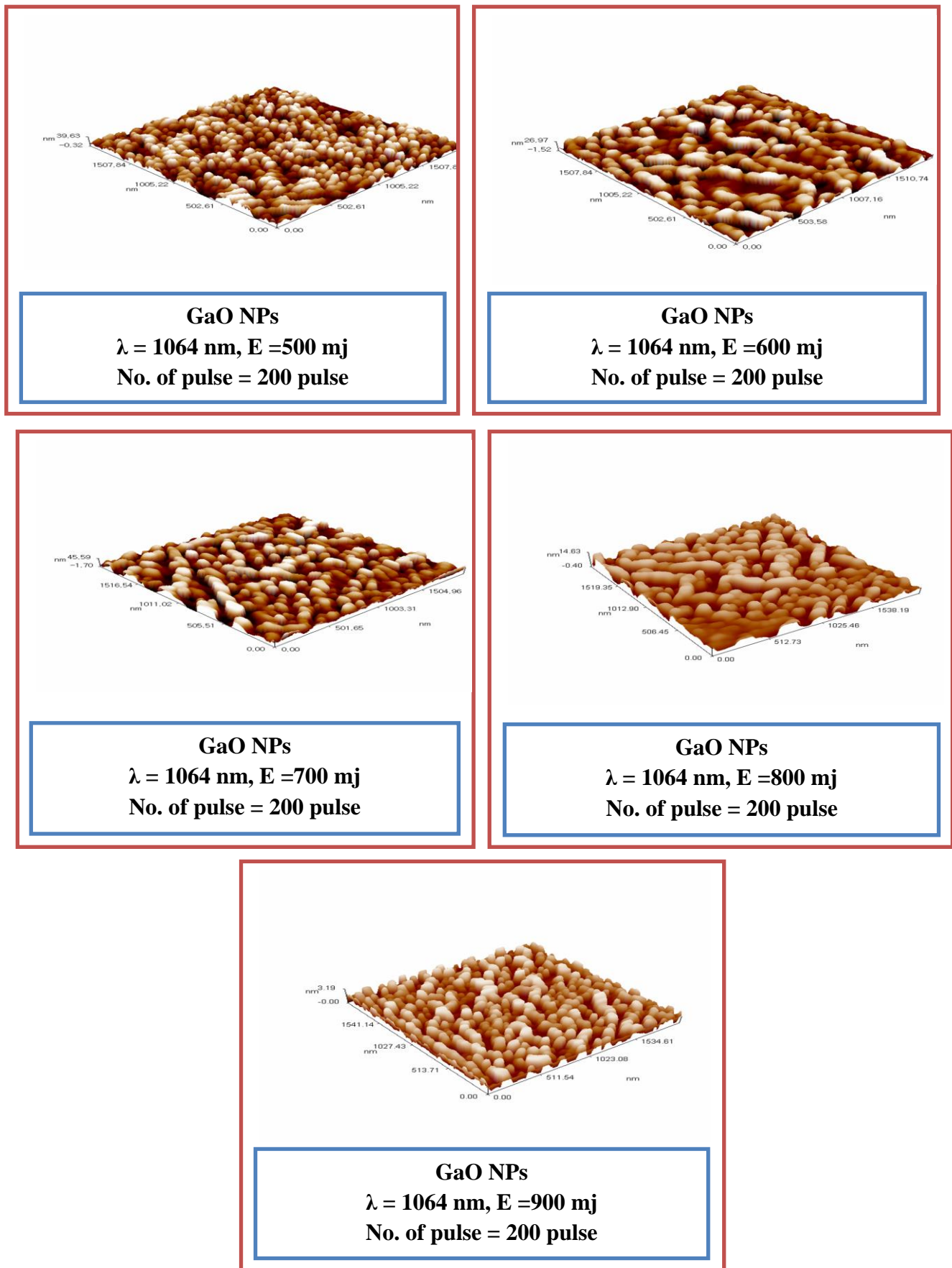


Figure 2- Images for surface morphology of Ga₂O₃ thin films analyzed by AFM at room temperature.

3.3. Electrical properties and optical measurements

In general, the nominally undoped Ga_2O_3 is extremely resistive due to the WBG [14, 15]. Deviations from oxygen stoichiometry, particularly oxygen vacancies, are invoked to explain the relatively high conductivity of n-type observed in certain nominally undoped Ga_2O_3 bulk crystals and thin films. This theory seems to be controversial, however, as hybrid functional density calculations indicate that oxygen vacancy is a deep donor with ionizing energy greater than 1 eV [16].

Because of the significance of Ga_2O_3 electrical transport properties, a detailed study of these properties was performed.

Figure 3-a shows the optical transmittance versus wavelength values of the thin films. The curves display a well-defined fringe pattern of interference, showing the thin films' smooth surface. The optical absorbance that was studied in the wavelength range between 190 to 1100 nm is illustrated in Figure- 3-b.

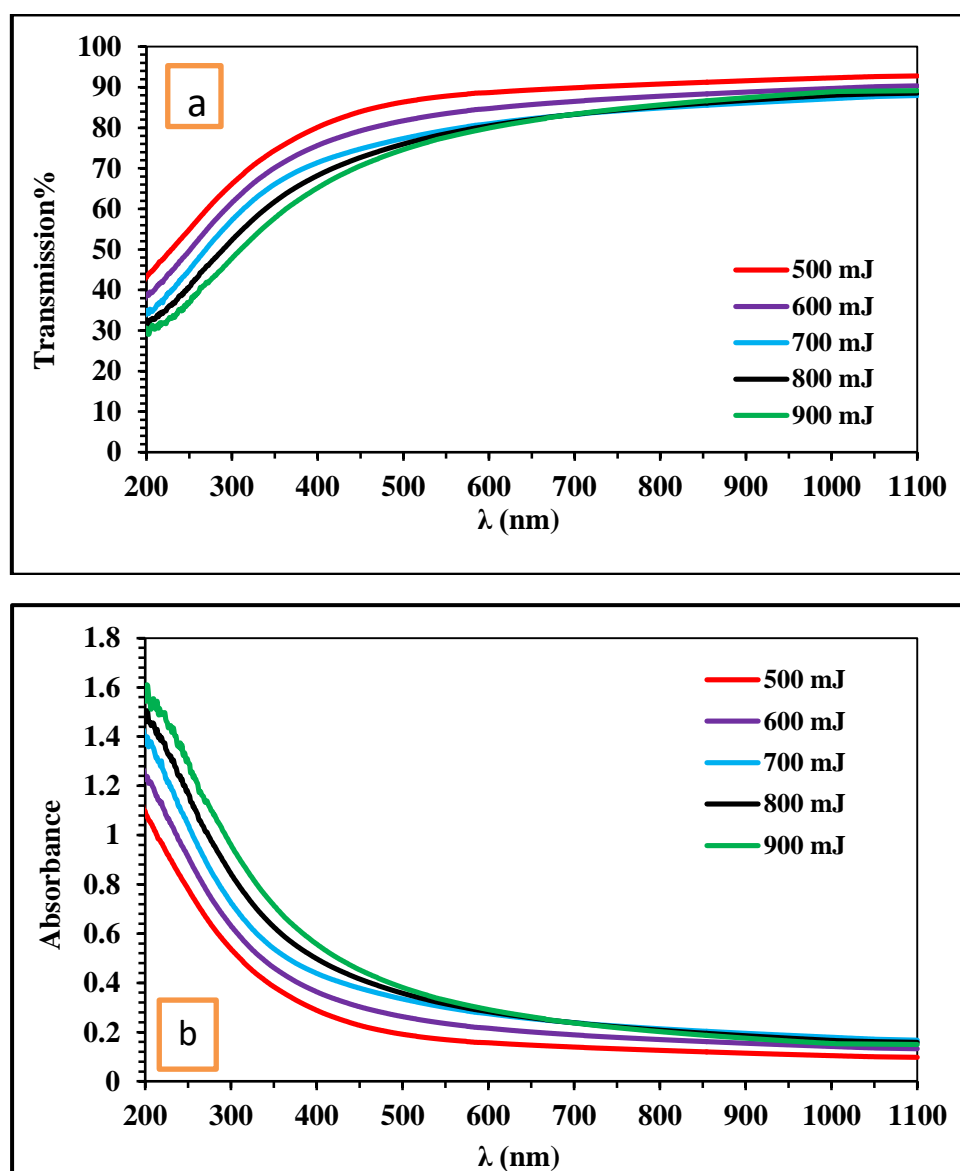


Figure 3-a and b: The optical transmittance and absorbance, respectively, of Ga_2O_3 thin films

As the aspect ratio of crystallites varies with the thickness, the optical reflection can be changed due to the morphological difference in films. The optical absorbance in the wavelength range of 450–800 nm ranged from 0 % to 10 %. The absorption was higher than 90% in the UV zone. Under the

energy band gap, the state of defects induces the absorption. The increase in the thickness of film will move the absorption edge towards longer wavelengths indicating a decrease in the band gap. Coefficient α can be calculated from the transmittance and absorption data, by using Lambert's formula [17]:

$$\alpha = \frac{1}{t} \ln\left[\frac{1}{T}\right] \dots\dots\dots (5)$$

where t is the thin film thickness, while the optical transmittance values are represented by T . For thin films, the optical band gap is determined using the model of Tauc and Davis as well as the model of Mott in the high absorption zone [17]:

$$\alpha hv = D (hv - E_g)^n \dots \dots\dots (6)$$

where the energy of the photon is expressed as hv , while E_g is the mean of the gap of the optical band and D is a constant. For $n = \frac{1}{2}$, the transition data provide the best linear fit in the band-edge field, which implies that the transition is of a natural course. Figure- 4. shows the method of measuring the band gap performed by using Tauc's plot and plotting the linear region in the graph to $(\alpha hv)^2 = 0$. The linear dependence of $(\alpha hv)^2$ on hv shows that the thin films of Ga₂O₃ are semi-conductors of direct transition forms. The thin film optical band gap was between 3.9eV and 4.4eV. It was found that the band gap is increasing with the increase in the film's thickness.

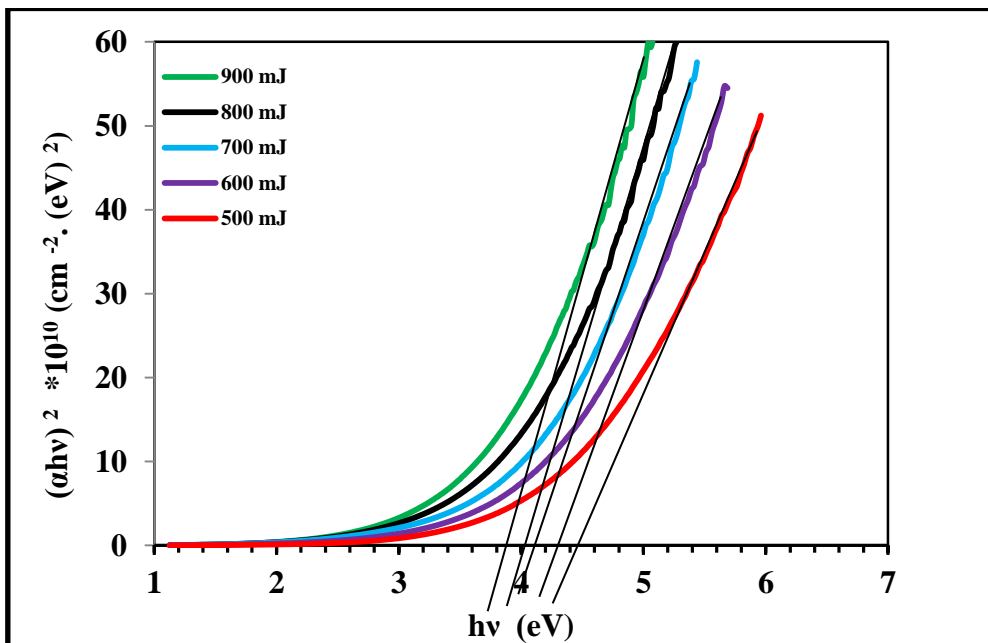


Figure 4- Energy band gap of Ga₂O₃ thin film

4. Conclusions

This study concluded that, as previously explained, different thicknesses of Ga₂O₃ thin films were deposited successfully on glass substrates. The deposited thin films were of single phases along the c -axis with the superior rise thin (111) plane. A compressive stress was generated due to the deposition parameters, such as growth temperature, gas pressure, laser energy and pulse period. The grains were grown uniformly on the film surface that was accurately described from AFM images in the surface topography study. Despite low average values and rams roughness, the texture of thin films is relatively smooth. The deposited thin films confirmed remarkable optical properties, including very high optical transmittance in the visible zone, low optical reflection, and a roughly sharp edge of absorption at all wavelengths. As shown above, the band gap could be efficiently reduced by Ga₂O₃. This study clearly showed high electrical conductivities in the thin films.

References

1. Rossi, M., Cubadda, F., Dini, L., Terranova, M. L., Aureli, F., Sorbo, A. and Passeri, D. **2014**. Scientific basis of nanotechnology, implications for the food sector and future trends. *Trends in Food Science & Technology*, **40**(2): 127-148.
2. Dey, Ananya. **2018**. "Semiconductor metal oxide gas sensors: A review." *Materials Science and Engineering: B* **229**: 206-217.
3. Po-Wei Chen, Shiau-Yuan Huang, Chao-Chun Wang, Shuo-Huang Yuan, Dong-Sing Wu, Influence of oxygen on sputtering of aluminum-gallium oxide films for deep-ultraviolet detector applications, *Journal of Alloys and Compounds*, **791**(2019): 1213e1219.
4. Guo, D.Y. Wu, Z.P. Li, P.G. An, Y.H. Liu, H. Guo, X.C. Yan, H. Wang, G.F. Sun, C.L. Li, L.H. Tang, W.H. **2014**. Fabrication of b-Ga₂O₃ thin films and solar-blind photodetectors by laser MBE technology, *Opt. Mater. Express*, **4** (2014): 1067e1076. <https://doi.org/10.1364/OME.4.001067>.
5. W.Y. Kong, G.A. Wu, K.Y. Wang, T.F. Zhang, Y.F. Zou, D.D. Wang, L.B. Luo, Graphene-b-Ga₂O₃ heterojunction for highly sensitive deep UV photodetector application, *Adv. Mater.* **28** (2016): 10725e10731. <https://doi.org/10.1002/adma.201604049>.
6. Lee, S. D., Ito, Y., Kaneko, K. and Fujita, S. **2015**. Enhanced thermal stability of alpha gallium oxide films supported by aluminum doping. *Japanese Journal of Applied Physics*, **54**(3): 030301.
7. Stepanov, S.I., Nikolaev, V.I., Bougrov, V.E. and Romanov, A.E. **2016**. GALLIUM OXIDE: PROPERTIES AND Applications, *Rev. Adv. Mater. Sci.* **44**(2016): 63-86.
8. Mortezaali, A., Taheri, O. and Hosseini, Z. S. **2016**. Thickness effect of nanostructured ZnO thin films prepared by spray method on structural, morphological and optical properties. *Microelectronic Engineering*, **151**: 19-23.
9. Kaur, Gurpreet, AnirbanMitra, and Yadav, K.L. **2015**. "Pulsed laser deposited Al-doped ZnO thin films for optical applications." *Progress in Natural Science: Materials International*, **25**(1): 12-21.
10. Aly, K. A., Khalil, N. M., Algamal, Y. and Saleem, Q. M. **2017**. Estimation of lattice strain for zirconia nano-particles based on Williamson-Hall analysis. *Materials Chemistry and Physics*, **193**: 182-188.
11. Meva, F. E. A., Ntomba, A. A., Kedi, P. B. E., Tchoumbi, E., Schmitz, A., Schmolke, L. and Lehman, L. G. **2019**. Silver and palladium nanoparticles produced using a plant extract as reducing agent, stabilized with an ionic liquid: sizing by X-ray powder diffraction and dynamic light scattering. *Journal of Materials Research and Technology*, (2019), 1-10.
12. Goyal, A., Yadav, B.S., Thakur, O.P., Kapoor, A.K. and Muralidharan, R. **2014**. Effect of annealing on b-Ga₂O₃ film grown by pulsed laser deposition technique, *J. Alloys Compd.* **583**: 214e219.
13. Chikoidze, E., Fellous, A., Perez-Tomas, A., Sauthier, G., Tchelidze, T., Ton-That, C. and Berini, B. **2017**. P-type β -gallium oxide: a new perspective for power and optoelectronic devices. *Materials Today Physics*, **3**: 118-126.
14. Stepanov, S.I., Nikolaev, V.I., Bourgov, V.E. and Romanov, A.E. **2016**. Gallium Oxide: properties and applications e a review, *Rev. Adv. Mater. Sci.* **44**: 63.
15. Grundmann, M., Klufel, F., Karsthof, R., Schlupp, P., Schein, F.-L., Splith, D., Yang, C., Bitter, S., von H. **2016**. Wenckstern, Oxide bipolar electronics: materials, devices and circuits, *J. Phys. D: Appl. Phys.* **49**: 213001.
16. Wang, Z., Nayak, P.K., Caraveo-Frescas, J.A. and Husam N. **2016**. Alshareef, recent developments in p-type oxide semiconductor materials and devices, *Adv. Mater.* **28**: 3831e3892.
17. Gurpreet Kaurn, AnirbanMitra, K.L.Yadav, Pulsed laser deposited Al-doped ZnO thin films for optical applications, *Progress in Natural Science: Materials International*, **25**(2015): 12–21.