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Iraqi Journal of Science, 2020, Vol. 61, No. 8, pp: 1952-1959 DOI: 10.24996/ijs.2020.61.8.12





ISSN: 0067-2904

Electroluminescence of Light- Emitting Organic Semiconductor / Europium Oxide Nanoparticle Hybrid Junction

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Received: 22/7/2019

Accepted: 19/11/2019

Abstract

A hybrid nanoparticles light emitting diode (NPs-LED) was fabricated as layers of ITO/TPD:PMMA/ Eu_2O_3 / Alq_3 / Al, by phase segregation method using spin coating technique. The NPs-LED hybrid device emitted light and consisted of three layers in a definite order placed on the transparent conducting oxide as an ITO substrate; the first layer was made of (N, N'-bis (3-methylphenyl) -N, N'-bis (phenyl) benzidine) (TPD) and polymethyl methacrylate (PMMA) polymers combined together. The second layer consisted of Europium (III) oxide (Eu_2O_3), while the third layer was Alq₃, one of the most frequently-used electron transport layers.

The electroluminescence (EL) of NPs-LED was studied with different bias voltages (20, 25 and 30V) at room temperature. Depending on the CIE 1931 color spaces, X the white light was generated at 30V and the green light at 20V. This was achieved by benefitting from the transition between deep levels in the energy gap of Eu₂O₃ NPs (surface state) and magnetic dipole transition for Eu⁺ (${}^{5}D_{0.3}and{}^{5}L_{6}$ to ${}^{7}F_{0.6}$). The Current – Voltage (I-V) characteristics demonstrated that the current varies with voltage and that the knee voltage value is 5 V. The EL spectrum showed a broad band emission, with a range of 350 - 700 nm. Finally, the correlated color temperature (CCT) was found to be about 7100 to 11500k.

Keywords: Europium oxide, Nanoparticle, electroluminescence, light-emitting diode

الإضائية الكهربائية لضوء المنبعث من شبه الموصل /دقائق اوكسيد اليوريبيوم النانوي لمفرق هجين

حسين رياض محمد ، عمر عدنان إبراهيم قسم الفيزياء ، كلية العلوم ، جامعة بغداد ، بغداد ، العراق

الخلاصة:

تم تصنيع الصمام الثنائي الباعث للجسيمات النانوية (NPs-LED) كطبقات من: : :ITO / TPD: تم تصنيع الصمام الثنائي الباعث للجسيمات النانوية (NPs-LED) كطبقات من: : :ITO / Eu2O3 / Alq3 / Al الدوران. يتكون الجهاز الهجين NPs-LED الذي ينبعث منه الضوء ، من ثلاث طبقات بترتيب محدد يوضع على أكسيد التوصيل الشفاف كركيزة ITO ؛ الطبقة الأولى ، كان البوليمر (N، -3) N'-bis (openyl) الم PN - Normethylphenyl والبوليمرات المعادين مع بعضهما, تمزج معهما الجزيئات النانوية من أكسيد استخداما. Europium مروجين مع بعضهما, تمزج معهما الجزيئات النانوية من أكمي استخداما.

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تمت دراسة التألق الكهربي (EL) الخاص بالـ NPs-LED لدرجات الجهد المختلفة للتحيز (20 ، 25 ، 30 فولت) في درجة حرارة الغرفة ، بدءًا من مساحة اللون 1931 CIE ، مما أدى إلى توليد الضوء الأبيض عند 30 فولت وتوليد الضوء الأخضر في 200. وذلك عن طريق الاستفادة من الانتقال بين المستويات العميقة في فجوة الطاقة لـ جزيئات النانوية لاوكسيد اليريبيوم Eu₂O₃ (الحالة السطحية) والانتقال تثائي القطب المعناطيسي لـ Eu₂O₃ النانوية لاوكسيد اليريبيوم Eu₂O₃ (الحالة السطحية) والانتقال تثائي القطب المعناطيسي لـ Eu₂O₃ النانوية لاوكسيد اليريبيوم 20³, والحالة السطحية) والانتقال تثائي القطب معناطيسي لـ Eu₂O₃ العاقة لـ جزيئات النانوية لاوكسيد اليريبيوم 20³, والحالة السطحية) والانتقال تثائي القطب المعناطيسي لـ Eu₂O₃ العاقبة العالي العربيوم 20³, والحالة السطحية) والانتقال تثائي القطب المعناطيسي لـ 2003 معالية النوية لاوكسيد اليريبيوم 20³, والحالة السطحية) والانتقال تثائي القطب المعناطيسي لـ 2003 معالية النوية لاوكسيد اليريبيوم 200³, والحالة السطحية) والانتقال تثائي العابي المعناطيسي المعناطيسي لـ 2003 معالية الوكسيد اليريبيوم 2003، والحالة السطحية التوار الناتيج جيد مقارنة بالفولتية وأن جهد العتبة التحيز 5 فولت. يُظهر الطيف Eu ناق واسع للانبعاث ، يتراوح بين معارية بالفولتية وأن جهد العتبة التحيز على درجة حرارة اللون المرتبطة (CCT) حوالي (7100 إلى 11500).

Introduction

Europium (III) Oxide (Eu₂O₃) is a synthetic compound made up of europium and oxygen. It is generally utilized as a red or blue luminescent substance in TVs and fluorescent lights because of its wide band gap for bulk (3.7eV) [1]. The deep sub levels (${}^{5}D_{0} \rightarrow {}^{7}F_{0, 1, 2, 3 \text{ and } 4}$) in Eu₂O₃ are resulting from the splitting of spectrum lines, which is known as the effect of stark on the atom. It can also partially occur because of the presence under the influence of an external electric field, which can be responsible for the emission of a band with different colors that include the whole visible spectrum [2, 3]. Due to this property, Eu₂O₃ is used in white LEDs [4]. The blend of nanoparticles of europium oxide (n-type) with some organic and inorganic semiconductors (p-type) will produce unique electronic and optical properties of the manufactured heterostructure device [5].

The result of the integration of semiconductor nanoparticles into organic matrices is the formation of a pn junction (EL device). Platting dielectric nanoparticles in an organic light-emitting hybrid junction affects the luminescence spectra, since the energy levels of the lowest unoccupied molecular orbital (LUMO) and the highest occupied molecular orbital (HOMO) of the TPD polymer are close to the energy level of the embedded semiconductor. Therefore, energy transfer will occur, as well as electronic polarization of the environment. If dielectric inclusions have a stronger polarization than the polymer matrix, the EL spectrum of the dielectric / polymer hybrid film will have red shifts compared with the fluorescence of the same dielectric material [6]. In this work, Eu_2O_3 NPs were used to make a TPD: PMMA: Eu_2O_3 NP hybrid coupler using spin coating and their EL and PL properties were studied. The light-emitting layers of the Eu_2O_3 NPs were obtained using the phase segregation method. This method is an inexpensive and simple solution process without using sophisticated vacuum equipment to manufacture a device based on Eu_2O_3 [1]

Experimental Work

For the preparation of hybrid films made of organic and inorganic materials, a transparent glass plate of indium oxide and tin (ITO) was used as the basis on which the layers of the thin films are located. The N, N'- diphenyl-N, N'- bis (3-methylphenyl) - 1,1'- biphenyl-4,4'- diamine (TPD) solution was prepared by dissolving 70 mg of the TPD in 1 ml chloroform. Likewise, a solution of PMMA in chloroform was prepared in a proportion of 3 g: 100 ml.. The two solutions of TPD and PMMA were mixed to prepare a blend with NPs in a ratio of 1: 0.5%. Eu₂O₃ was added at a percentage of X / 200 or X * 0.005 to the blend of TPD:PMMA. The mixture of blend and NPs was prepared by a magnetic stirrer and an ultrasound path, after that the film was prepared. We used a clean ITO, on which the mixture (blend and Eu_2O_3) was deposited by spin-coating at 2500 rpm for 30 seconds. This was followed by drying by oven at 70 ° C for 2 hours to obtain the first layer of the device. In the same time, Tris (8-hydroxyquinolinato) aluminum (Alq₃) (electron injection layer) was dissolved in ethanol (insoluble solvent of TPD) at a ration of 70 mg: 3 ml. In the same way, this layer was deposited on the first layer by spin-coating at 2500 rpm for 30 seconds. Then the films were placed in oven at 60 $^{\circ}$ C for 2 hours to eliminate the remaining solvents. Finally, aluminum contact was deposited by thermal evaporation under 4*10⁻⁵ mbar pressure as a negative cathode to obtain the (ITO/TPD: PMMA/ Eu₂O₃/Alq₃/Al) device .



Figure 1-Structure of hybrid junction devices using Eu₂O₃NP₈

Figure-1 demonstrates the structure of the hybrid junction (TPD:PMMA/Eu₂O₃ NPs) device prepared by the phase segregation of Eu₂O₃ nanoparticles on the organic blend layer. The structure of the device is comprised of three layers kept progressively on the base layer of ITO by phase segregation through utilizing the spin-coating technique. Indium tin oxide (ITO) is the most generally utilized option to incorporate other straightforward conductive oxides. The natural layer comprised of the TPD is a hole transport film, while the Eu2O3 NPs and Alq₃ films form the electron transport layer. The phase segregation method was used because the viscosity of NPs in chloroform is different from that of the blend (TPD: PMMA), which leads to separating them through the spin-coating. To investigate the electrical properties of the hybrid junction device, Hall Effect measurements were conducted by HMS-3000 and I-V characteristics were studied to assess the EL device by a Keithley 616 advanced electrometer and a DC source. The EL range of the EL device was estimated at room temperature by a fiber-optic spectrometer of the CCS arrangement (thorlabs, Germany) in the inclusion scope of 200 - 1000 nm and targets of 1 nm.

X-Ray Diffraction Results for Eu₂O₃ NPs

The crystal graphic orientation of Eu_2O_3 NPs was studied by X-ray diffraction patterns recorded at a scanning range (2 θ) of 5° – 80°, as shown in Figure-2. The X-ray diffraction patterns included many diffraction peaks. The high intensity of the orientation peak was at $2\theta = 27.1865^{\circ}$ which is diffracted from miller indices (222), while the crystal structure used was wurtzite. There were no crystalline defects in the X-ray. From the linear fit of the data, the crystal size (d) was found to be40.2 nm).



Figure 2-X-ray diffraction pattern of Eu₂O₃ NPs.

Electrical Measurements

The Hall Effect system type (Ecopia HMS -300) was used to study the electrical properties of Eu_2O_3 , for which the results are outlined in Table-1. The film of 1µm density indicates a semiconductor of n – type conductivity. Table-1, shows the results of conductivity and Hall Effect parameters of the Eu_2O_3 NPs that was deposited by drop casting on a glass substrate etched by HF solution.

Table 1-Conductivity	and Hall measuremen	nts of the Eu ₂ O ₃ NPs
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sample	Conductivity (Ω.cm) ⁻¹	Mobility (cm²/V.s)	Bulk Concentration (cm ³)	Hall Coefficient (m²/C)	Resistivity (Ω.cm)
Eu ₂ O ₃	1.584E-4	6.292E+2	-1.571E+12	1.584E-4	6.315E+3





Figure 3-Semi- log plot of the current-voltage (I-V) characteristics for ITO/TPD: PMMA/ $Eu_2O_3/Alq_3/Al hybrid junction device.$

The resulting I-V characteristics of the ITO/TPD: PMMA/ Eu₂O₃/Alq₃/Al hybrid junction device were analyzed using the standard thermionic emission theory [7]. The results illustrate that the current transport mechanism was exhibited in three different regions. The current density in region (I) (<5 V) pursues a direct linear relationship (Ohmic relationship) (I α V). This indicates that the current transport is dominated by tunneling at low voltages [8]. The boundary for this area was determined below the knee tension (5V).

In region (II) (5-20 V), the current density increased exponentially as (I α exp (V)) carriers are injected into the dielectric are produced because of a thermionic procedure coursing through the hindrance. Thus, the number of the free charges was less than the ratio of the trapped charges. The results also demonstrated the exponential voltage dependence [8]; with further utilization of voltage,

the injected carriers quickly increase, and the traps are filled (the traps states are ${}^{5}D_{0}$, ${}^{5}D_{1}$, ${}^{5}D_{2}$, ${}^{5}D_{3}$, ${}^{5}L_{6}$, NPs state and defect stated) [2] at 20V which is called Traps-Filled-Limit Voltage (VTFL).

Above 20V (region III), the current density obeys the power law $(I\alpha V^2)$. The light discharge starts at region (III), in 20V and 610µA, in light of the fact that the transport through the ITO/TPD: PMMA/ Eu₂O₃/Alq₃/Al The hybrid layer is driven by Trap-Charge Limited Current (TCLC) in the hole of the Eu₂O₃ NPs. This infers that the density of the injected charge carriers is significantly important as compared to the thermally delivered free charge carriers. Space-Charge Limited Current (SCLC) is the dominant charge carrier transport [9].

The I–V characteristics of the three samples show an exponential increase in current due to the decrease in the depletion layer width at the interface. Under the forward bias, the conduction band barrier will be decreased due to the exponential distribution of electrons and holes within the conduction and valence bands. Hence, the diffusion current flows across the junction increases exponentially with increasing the forward bias. The drift current flowing in the opposite direction does not depend on the potential barrier height and will enhance the electron flow from the n-(NPs) to the p-(TPD) and that of the holes from the p-(TPD) to the n-(NPs). The successive recombination would give rise to the forward bias current flow. This means that the diode resistance would firstly decrease upon the increase of the forward bias voltage (i.e., the lowering of the barrier). The semi-log plot of the I-V data at RT is shown in Figure- 3 and it illustrates that the current transport mechanism is exhibited in three different regions.

Electroluminesces Measurements

Electroluminescence estimations at forward bias voltages of 25 V speak to possible for the light which has been acquired tentatively through the hybrid device. Hybrid junction devices were carried out using a photomultiplier detector at room temperature. The transport component of the carrier in this hybrid coupling device performs so that the TPD works as a hole transport layer and Alq₃ as an electron transport layer. The gaps are brought by the ITO anode in the HOMO of the terminal potential difference grid. It is then shipped to the VB or to ${}^{7}F_{0}$, ${}^{7}F_{1-2}$, ${}^{7}F_{3-4}$, and ${}^{7}F_{5-6}$ of the Charge Transfer (CT) of Eu₂O₃. In the meantime, electrons are injected at the cathode into the Alq3 LOMO networks and moved to the conduction band (CB) or to ${}^{5}D_{0}$, ${}^{5}D_{2}$, ${}^{5}D_{3}$, and ${}^{5}L_{6}$. The condition of the CT and the demonstrated imperfection of the CT of Eu₂O₃ [2] are presented in Figure-5. Subsequently, holes and electrons in Eu₂O₃ NPs are also recombined as excisions , also known as inter band recombination. This recombination process of holes and electrons through the deformities leads to produce a light at various wavelengths. The process that involves the recombination by imperfections is called Shockley-Reed-Hall recombination [10, 11].



Figure 4-EL of TPD:PMMA/Eu₂O₃/Alq₃ at different voltages.



Figure 5-Energy bands diagram of ITO/TPD: PMMA/Eu₂O₃/Alq₃/Al hybrid device [2,3, 13and 14].

The violet radiation at \approx 360 nm is caused by the band to band transition [2]. The transition from the blue emission of 430 nm to blue emission of 513 nm corresponds to the transitions of ${}^{5}D_{3}$ and ${}^{5}D_{2}$ to ${}^{7}F_{0}$, ${}^{7}F_{2}$ and ${}^{7}F_{3}$. This result is in a god agreement with Yoshimoto et al. [12]. The EL peak at 410nm is corresponding to the emission of TPD because the energy of these peaks is equal to the band gap of TPD 3eV [6]. The peak at 410nm appeared when utilizing NPs of Europium oxide, where vitality utilizing Auger is expected to nanoscale substances that cause a high Auger recombination cross area. With an abatement in TPD, the number of injected holes through the interface NP TPD/ Eu₂O₃ is diminished. Electrons are productively injected into the conduction band of Eu₂O₃ NPs [13, 14].

Excitations through nan-radiative vitality move from the TPD to the Eu2O3 NPs because the EL device structures with a sub-10nm separation between TPD and Eu2O3 NPs [15]. The green emission 513- 585nm corresponds to ${}^{5}D_{0}$, ${}^{5}D_{1}$ and ${}^{5}D_{2}$ to ${}^{7}F_{0}$, ${}^{7}F_{1-2}$ and ${}^{7}F_{3}$ transitions [2, 12]. The orange 595nm to red emission around 700 nm is related to ${}^{5}D_{0}$ to ${}^{7}F_{0}$, ${}^{7}F_{1-2}$ and ${}^{7}F_{3-4}$ transitions or to the nan-radiative vitality of the surface state in nanostructure to V.B of Eu₂O₃ [3]. All of these advances are shown in Figure-5). The shade of light is situated by the color spaces CIE 1931 Chromaticity Coordinate (Figure-6). It implies that the x and y are obtained with the help of X Y & Z.

$$x = \frac{X}{X + Y + Z}$$
$$y = \frac{Y}{X + Y + Z}$$

where all the three coordinates are zone underneath bend for red, green and blue are districts in EL range, separately. The correlated color temperature(CCT) can be calculated with the help of McCamy's formula. The correlated color temperature with the use of x and y chromaticity is calculated as in Eq. 3 [13]:

$$CCT = -449n^3 + 3525n^2 - 6823n + 5520.33 - \cdots (3)$$

where n equals to, n = (x - 0.332)/(y - 0.1855)Table-2, shows the values of x, y and CCT.

Simple	Bias voltage	х,у	CCT(K)
ITO/TPD: PMMA/0.5% w Eu ₂ O ₃ /Alq ₃ /Al	20	0.3,0.327	7213 K
	25	0.168,0.43	11416 K
	30	0.306,0.29	7718 K

Table 2-CCT and x, y coordinates for different bias voltages.



Figure 6-CIE 1931 chromaticity diagram for ITO/TPD: PMMA/ Eu2O3/Alq3/Al hybrid junction device.

Conclusions

As a summary of the results, we demonstrated that organic - inorganic hybrid composites emit green and white light emission by controlling the bias voltage from the transition through the deep levels in energy gap of Eu_2O_3 NPs ($5D_{0-3}$ and $5L_6$ to $7F_{0-6}$) and through the defect states of these NPs. White light generation is attributed to the increased number of transitions through the deep levels of Eu_2O_3 by increasing the bias voltage . The bandwidth of the EL of the hybrid film spans as wide as 250nm. The white light emission is attributed to the interaction of the organic and inorganic components of the hybrid film.

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